

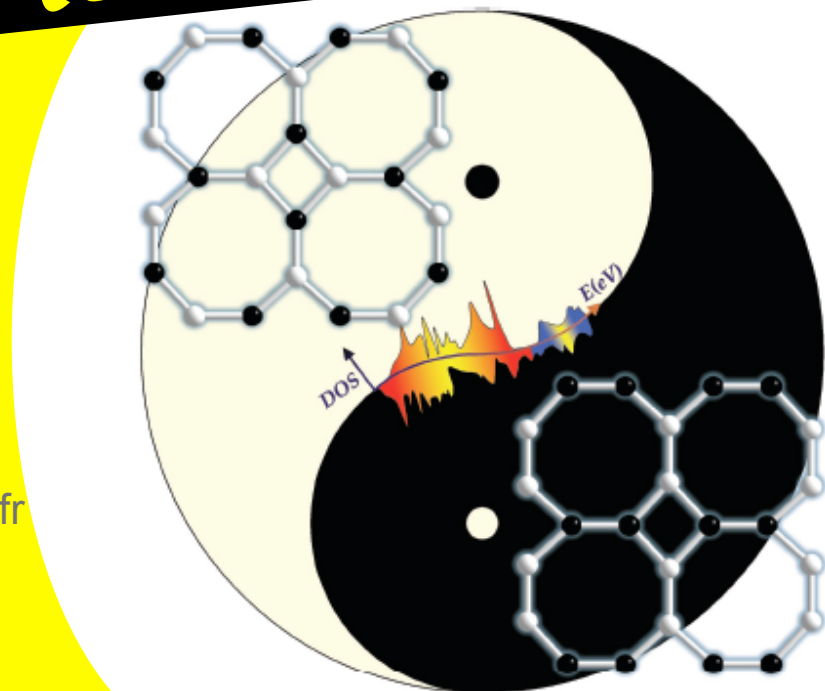
Discover how easy it is to
grasp the basics of band theory

Band Theory FOR DUMMIES

Jean-François Halet and Jean-Yves Saillard

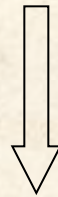
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Electronic Structure of Solid Materials: Basic Concepts

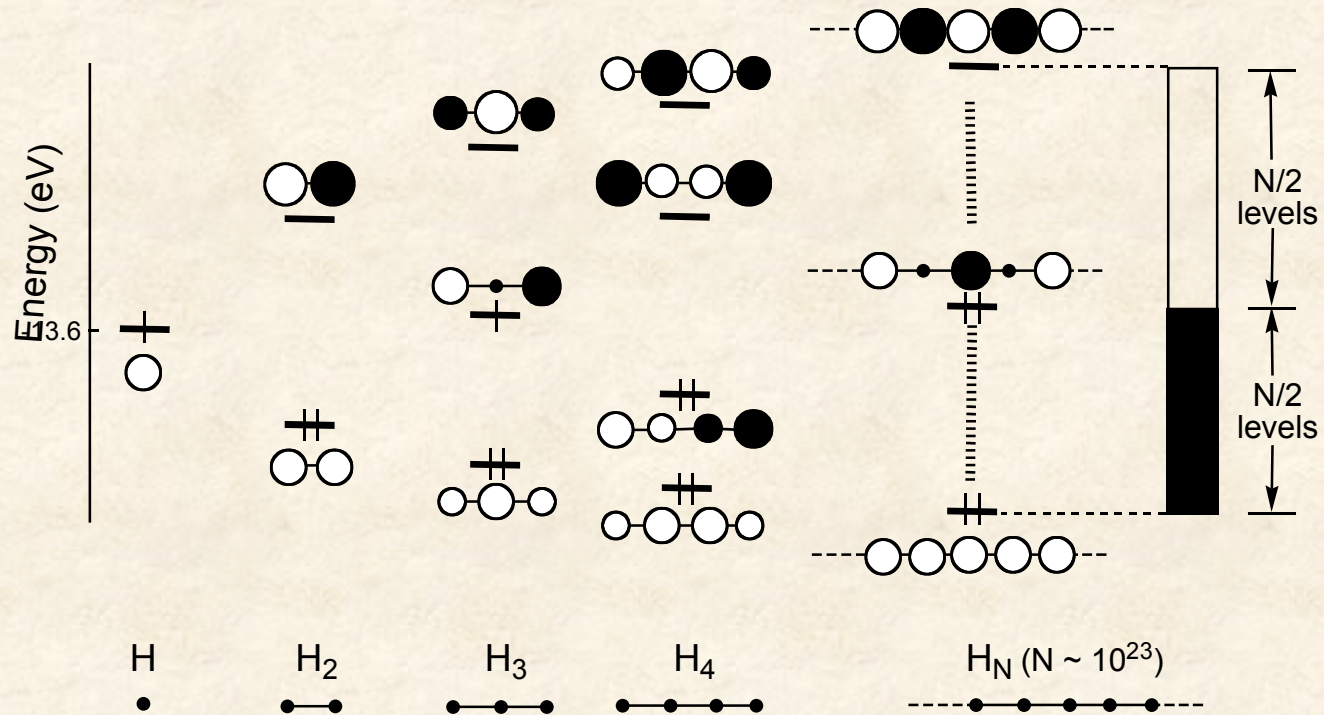
Solid materials \equiv Periodic solids



The electronic structure of solids (1D, 2D or 3D)
is usually discussed in terms of **band theory**

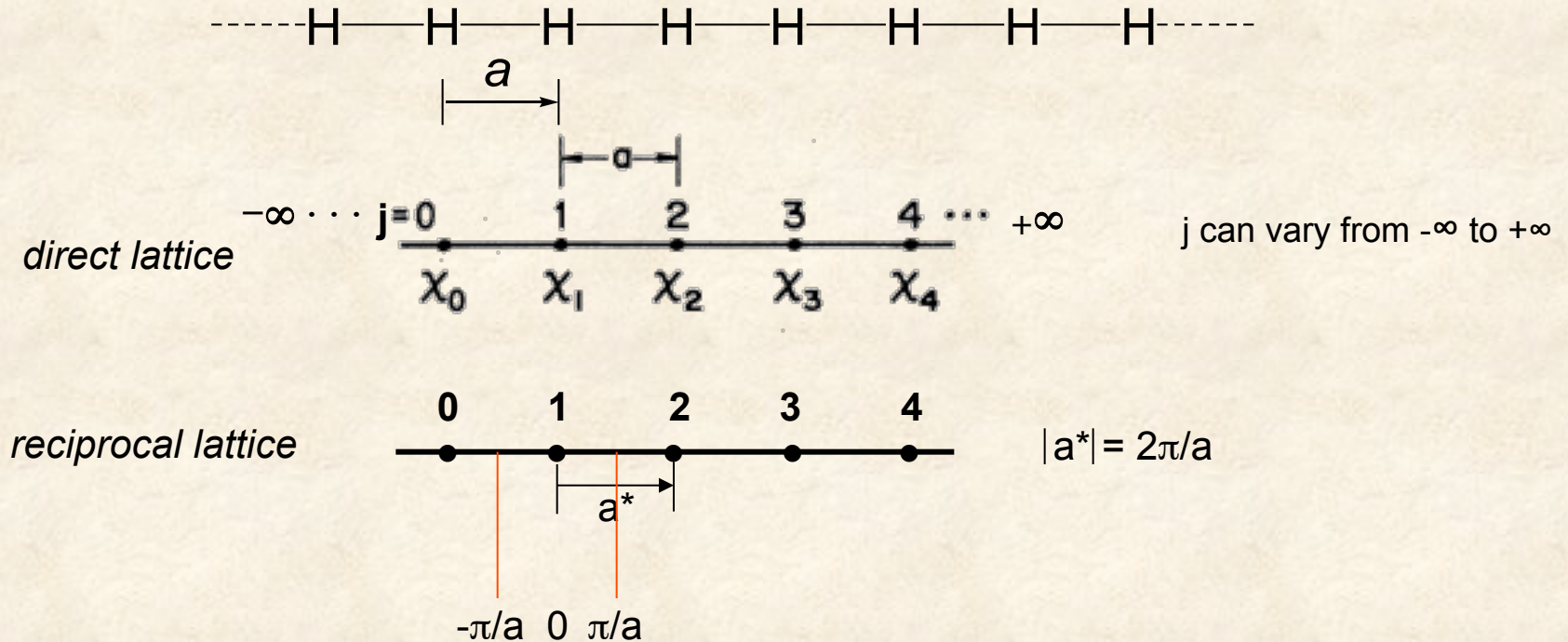
Orbitals and Bands in One Dimension

- The "molecular" approach: Linear H_n oligomers ($n = 1, N \sim 10^{23}$)



Bloch Functions, Crystal Orbitals and Band Structures

- The "solid state" approach: Infinite chain of equidistant H atoms



Bloch orbital (function):
$$\phi(k) \propto \sum_j [\exp(ikR_j) \cdot \chi_j(r - R_j)]$$

where: k = index (wave vector, vector belonging to the reciprocal lattice)

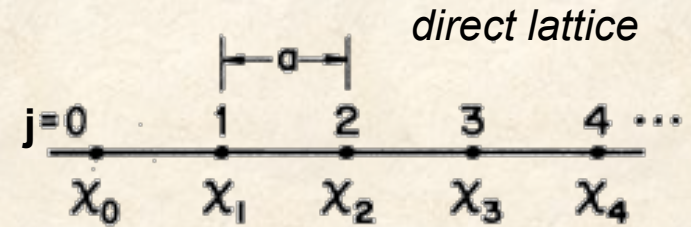
$R_j = ja$ (j , integer varying from $-\infty$ to $+\infty$)

$\chi_j = 1s(H_j)$

Bloch Functions, Crystal Orbitals and Band Structures

$$\phi(k) \propto \sum_j \exp(ikja) \chi_j$$

$$\phi(k) \longrightarrow E(k)$$



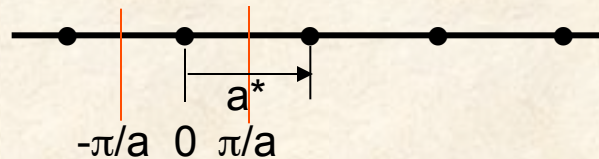
$$E(k) = f(k)$$

Periodical function in the reciprocal space (period = $2\pi/a$)

$\phi(k)$ and $\phi(-k)$ are degenerate, i.e. $E(k) = f(k)$ is a *centrosymmetrical function*

Thus, the study of the function $E(k) = f(k)$ can be restricted to the interval: $0 \leq k \leq \pi/a$
(Irreducible part of the first Brillouin zone)

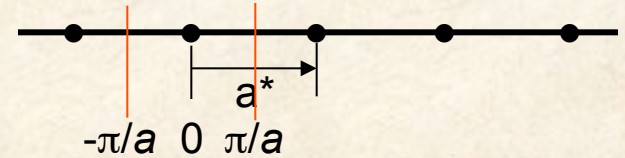
reciprocal lattice



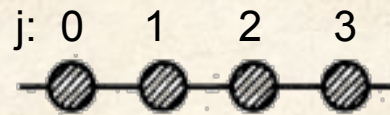
$$|a^*| = 2\pi/a$$

Bloch Functions, Crystal Orbitals and Band Structures

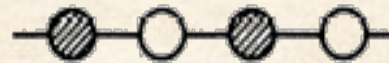
Bloch orbitals in the first Brillouin zone ($0 \leq k \leq \pi/a$)



$$k = 0 \quad \phi(0) \propto \sum_j \exp(ikja) \chi_j = \sum_j \exp(0) \chi_j = \sum_j \chi_j = \chi_0 + \chi_1 + \chi_2 + \chi_3 + \dots$$

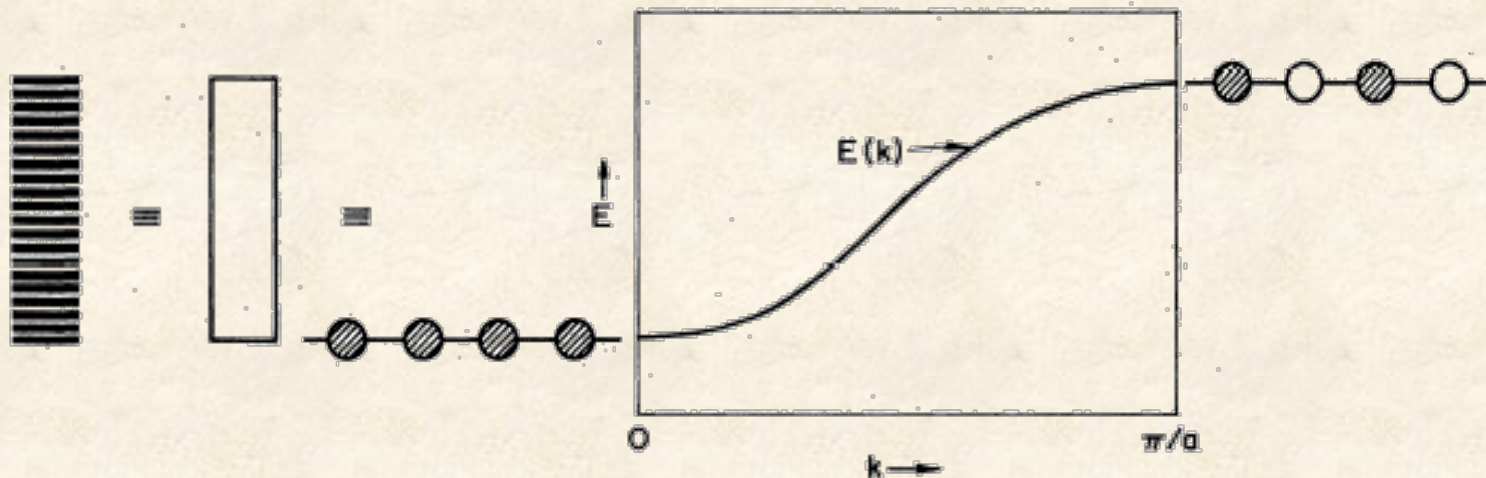


$$k = \pi/a \quad \phi(\pi/a) \propto \sum_j \exp(ikja) \chi_j = \sum_j \exp(i\pi j) \chi_j = \sum_j (-1)^j \chi_j = \chi_0 - \chi_1 + \chi_2 - \chi_3 + \dots$$

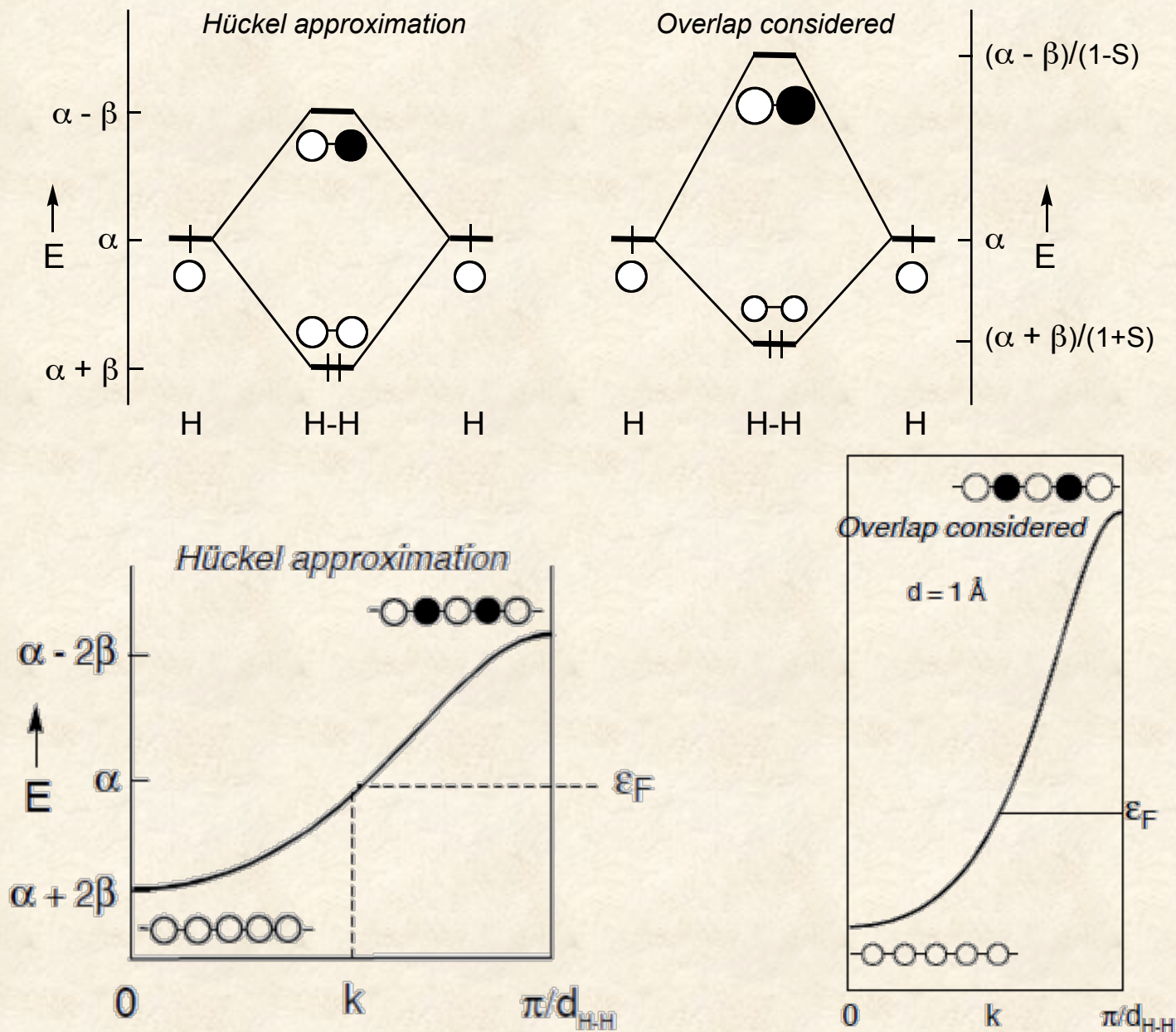


$\phi(k) \longrightarrow E(k)$ and $\phi(-k) \longrightarrow E(k)$

Band structure: $E(k) = f(k)$ for $0 \leq k \leq \pi/a$



Bloch Functions, Crystal Orbitals and Band Structures



Band Width

- Dispersive vs flat bands

The band width W depends on the overlap between neighboring atoms, *i.e.* on the interatomic distance a .

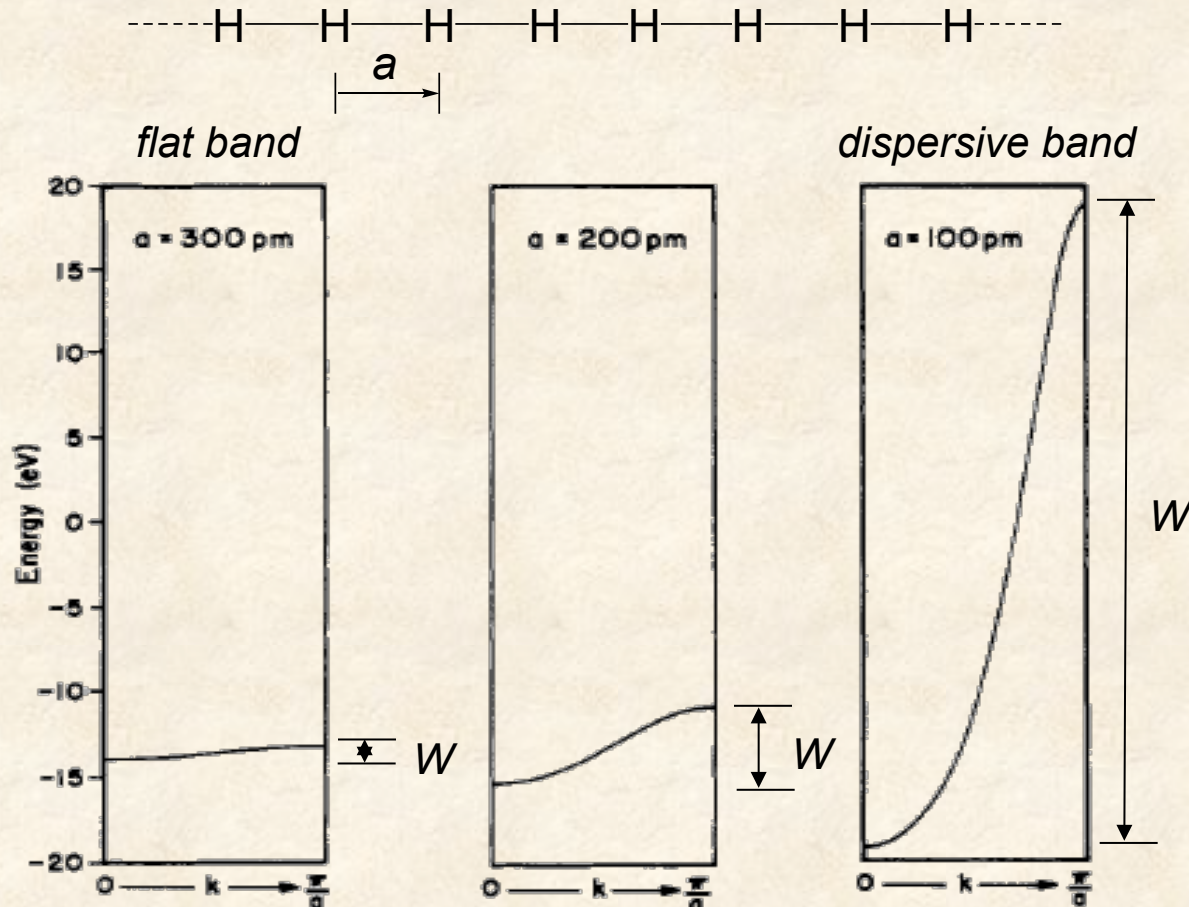
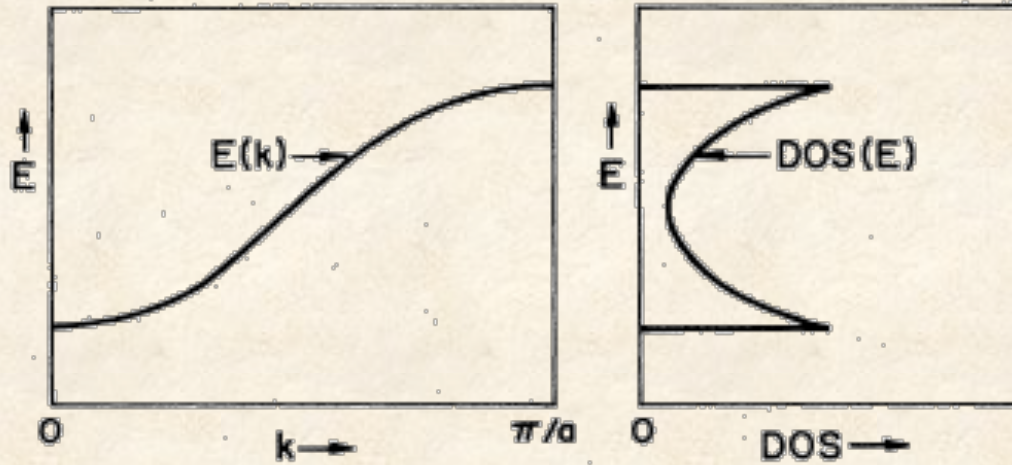


Figure 2. Band structure of a chain of hydrogen atoms spaced 300, 200, and 100 pm apart. The energy of an isolated H atom is -13.6 eV .

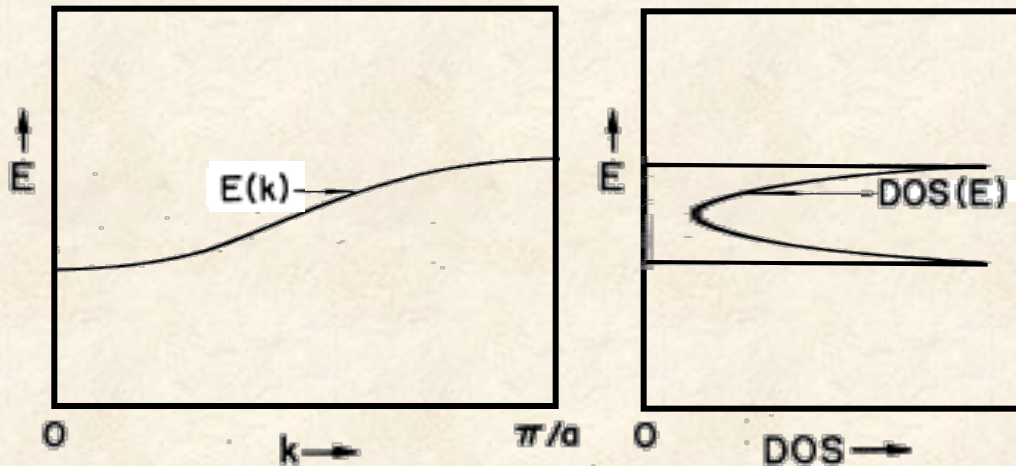
Density of States (DOS)

DOS (E)dE = Number of states (levels), n , between E and E + dE

$a = d_{H-H}$ short
Dispersive band



$a = d_{H-H}$ long
Flat band

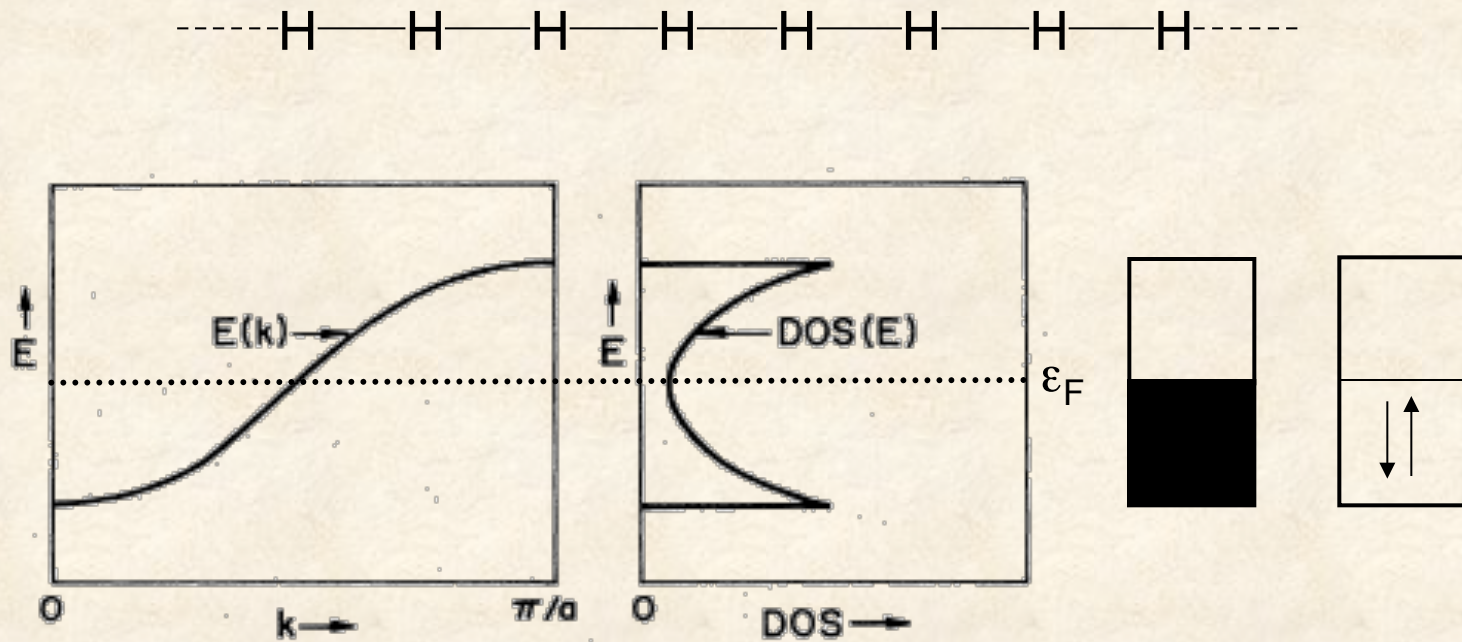


$$n(E) \propto (\delta E(k)/\delta k)^{-1}$$

NB: The area of the surface delimited by the curve $DOS(E) = f(E)$ is constant when a varies

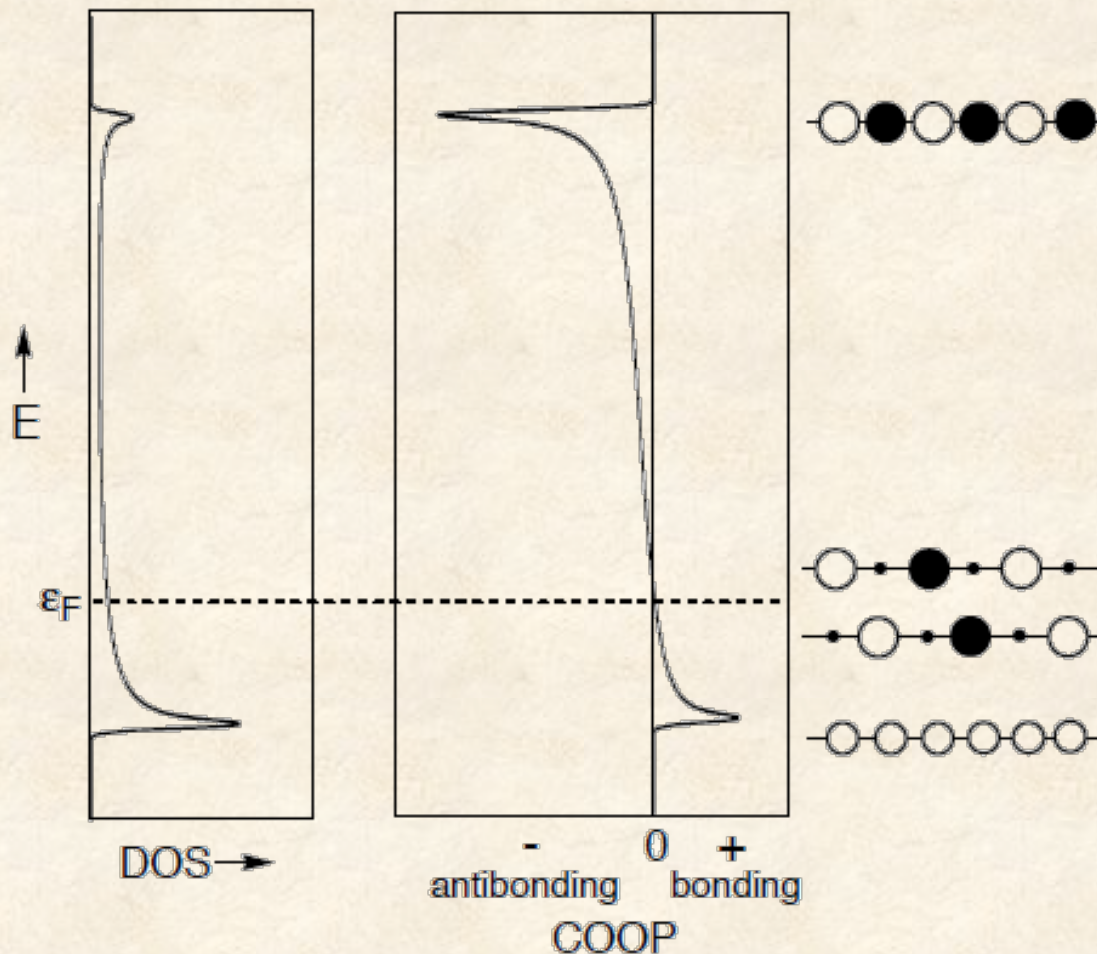
The Fermi Level (ϵ_F)

Fermi level (ϵ_F) (solid) \equiv HOMO (molecule)



The Crystal Orbital Overlap Population (COOP) curve

Mulliken overlap population vs. energy

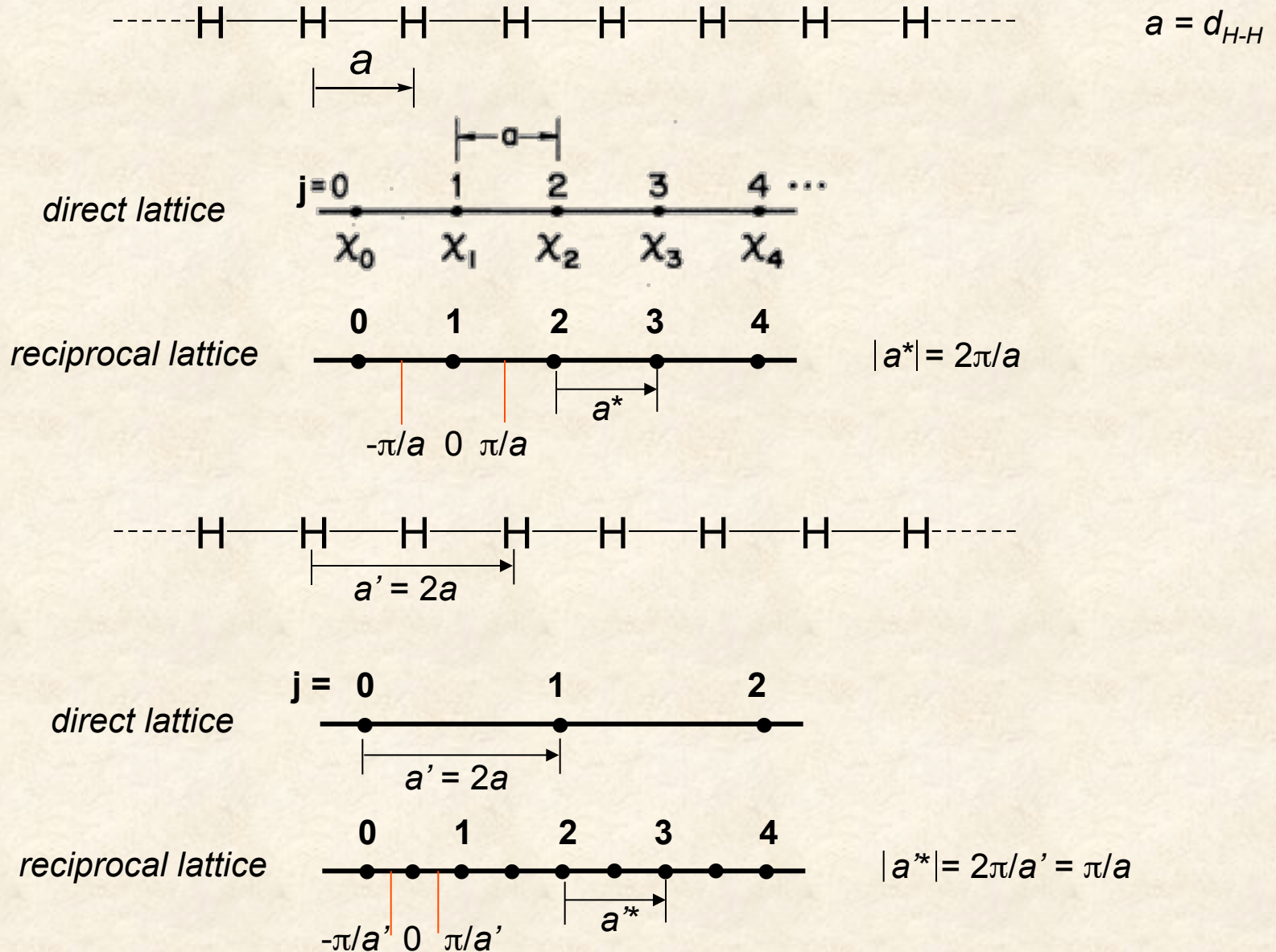


Mulliken overlap population:

$$\Psi = c_1\chi_1 + c_2\chi_2$$

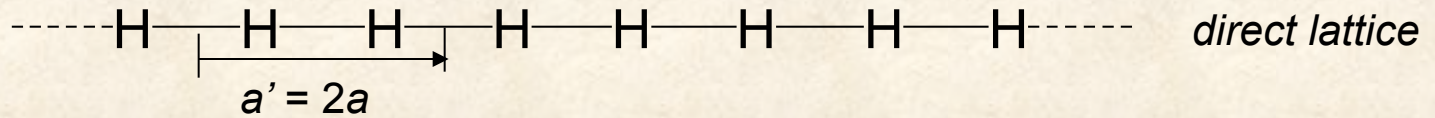
$$\text{ED (MO): } 1 = c_1^2 + c_2^2 + 2c_1c_2S_{12}$$

More than One Elementary Unit in the Unit Cell: Folding Bands



More than One Elementary Unit in the Unit Cell: Folding Bands

$$a = d_{H-H}$$



2 H per unit cell, thus 2 χ_j orbitals : χ_j^σ and $\chi_j^{\sigma^*}$



Bloch orbitals in the irreducible part of the first Brillouin zone ($0 \leq k \leq \pi/a'$):

$$\phi(k) \propto \sum_j [\exp(ikR_j) \cdot \chi_j], \text{ with } R_j = j \cdot a \text{ and } \chi_j = \chi_j^\sigma \text{ or } \chi_j^{\sigma^*}$$

$$k = 0 \quad \phi_\sigma(0) \propto \sum_j \exp(0) \chi_j^\sigma \text{ and } \phi_{\sigma^*}(0) \propto \sum_j \exp(0) \chi_j^{\sigma^*}$$

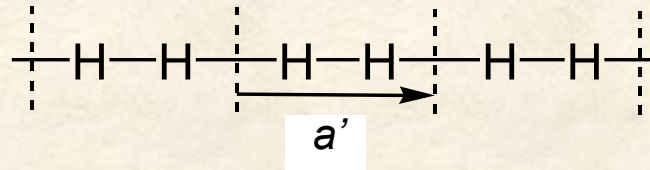


$$k = \pi/a' = \pi/2a \quad \phi_\sigma(\pi/a') \propto \sum_j \exp(i\pi j) \chi_j^\sigma \text{ and } \phi_{\sigma^*}(\pi/a') \propto \sum_j \exp(i\pi j) \chi_j^{\sigma^*}$$

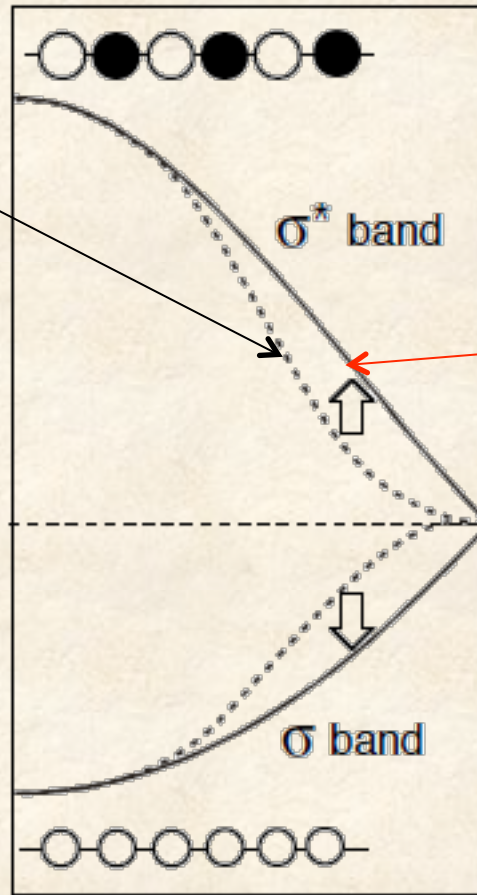
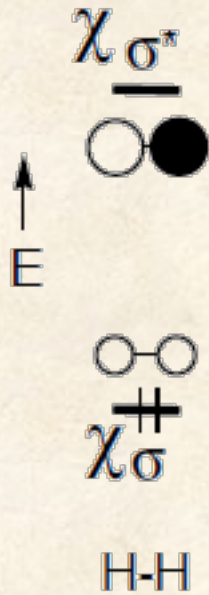


More than One Elementary Unit in the Unit Cell: Folding Bands

$$a' = 2d_{H-H}$$



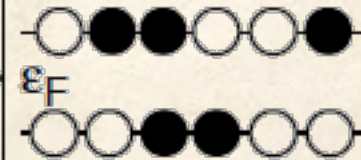
Bloch Orbital (BO)



$$H^{\text{eff}} \psi_i(k) = e_i(k) \psi_i(k)$$

$$\psi_i(k) = \sum_{\mu=1}^M C_{\mu i}(k) \phi_{\mu}(k)$$

Crystal Orbital (CO) =
Linear combination of BO



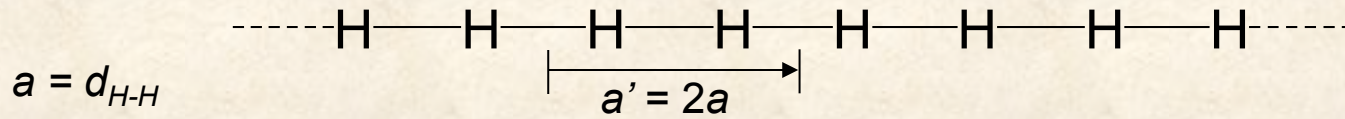
0

k

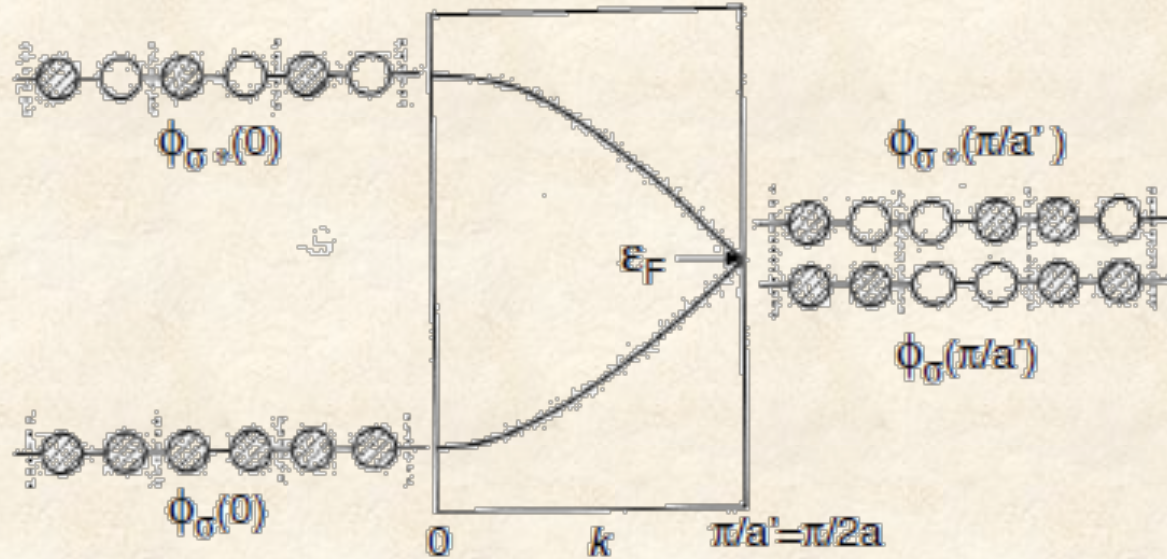
$\pi/a' = \pi/2a$

ϵ_F

More than One Elementary Unit in the Unit Cell: Folding Bands

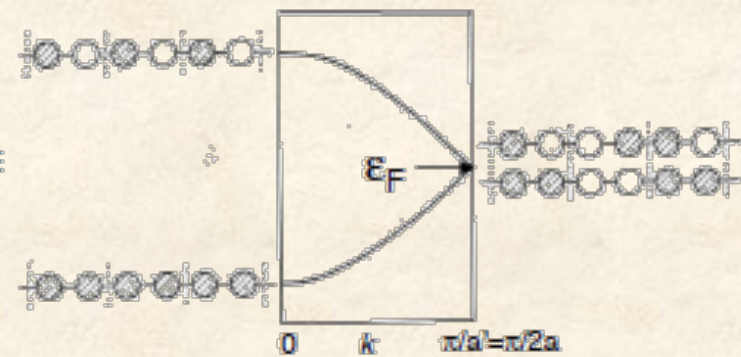
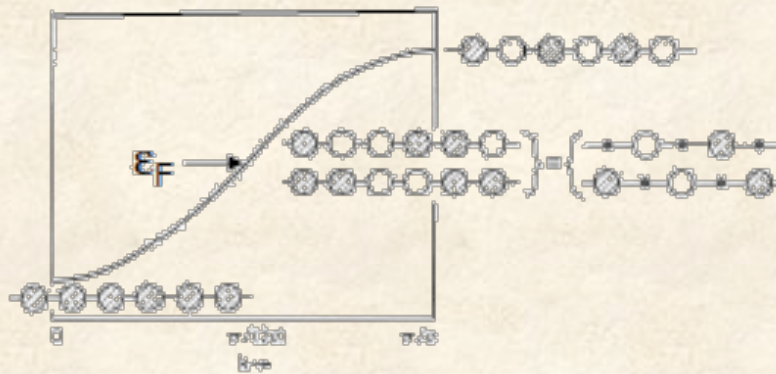
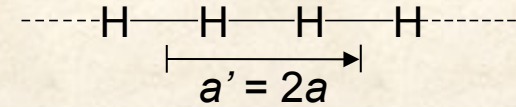
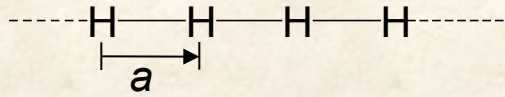


Band structure: $E(k) = f(k)$

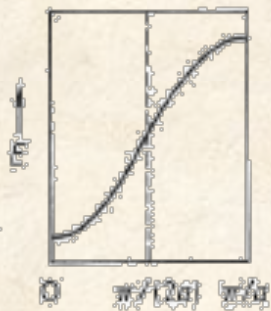


More than One Elementary Unit in the Unit Cell: Folding Bands

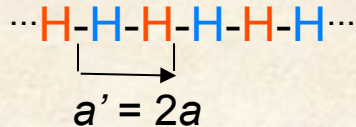
$$a = d_{H-H}$$



Band folding:



Exercise: Generate the band structure of the double cell ($a' = 2a$) by considering the interaction between two interpenetrating H networks



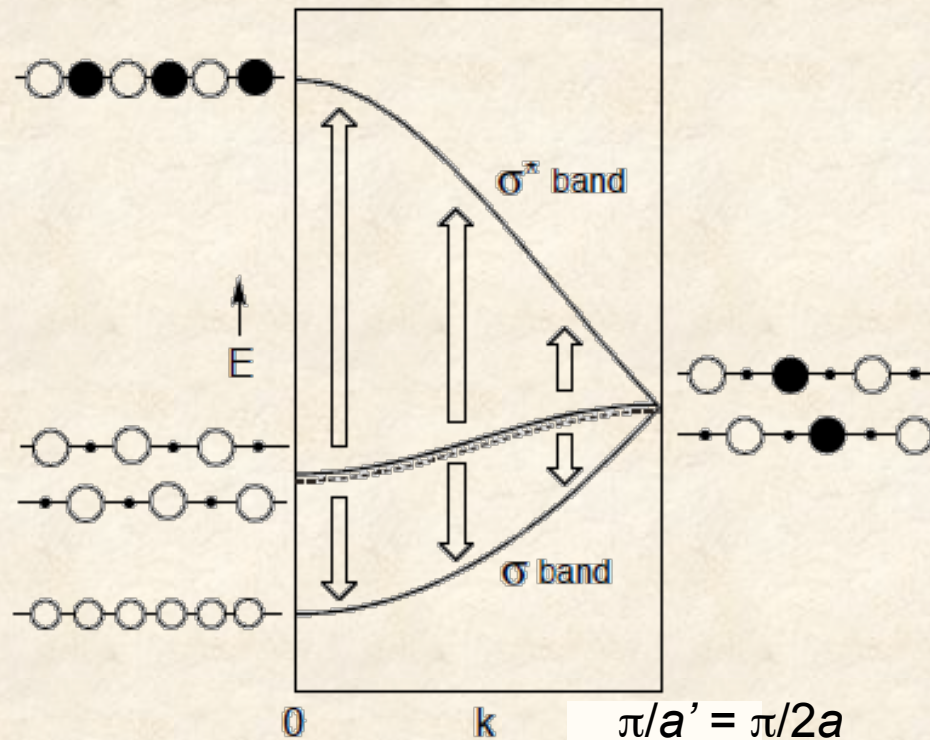
The $\cdots \text{H} \cdots \text{H} \cdots$ network generates a flat band (very weak overlap between red atoms).

Similarly, the $\cdots \text{H} \cdots \text{H} \cdots$ network generates a flat band.

Both flat bands interact and repel each other.

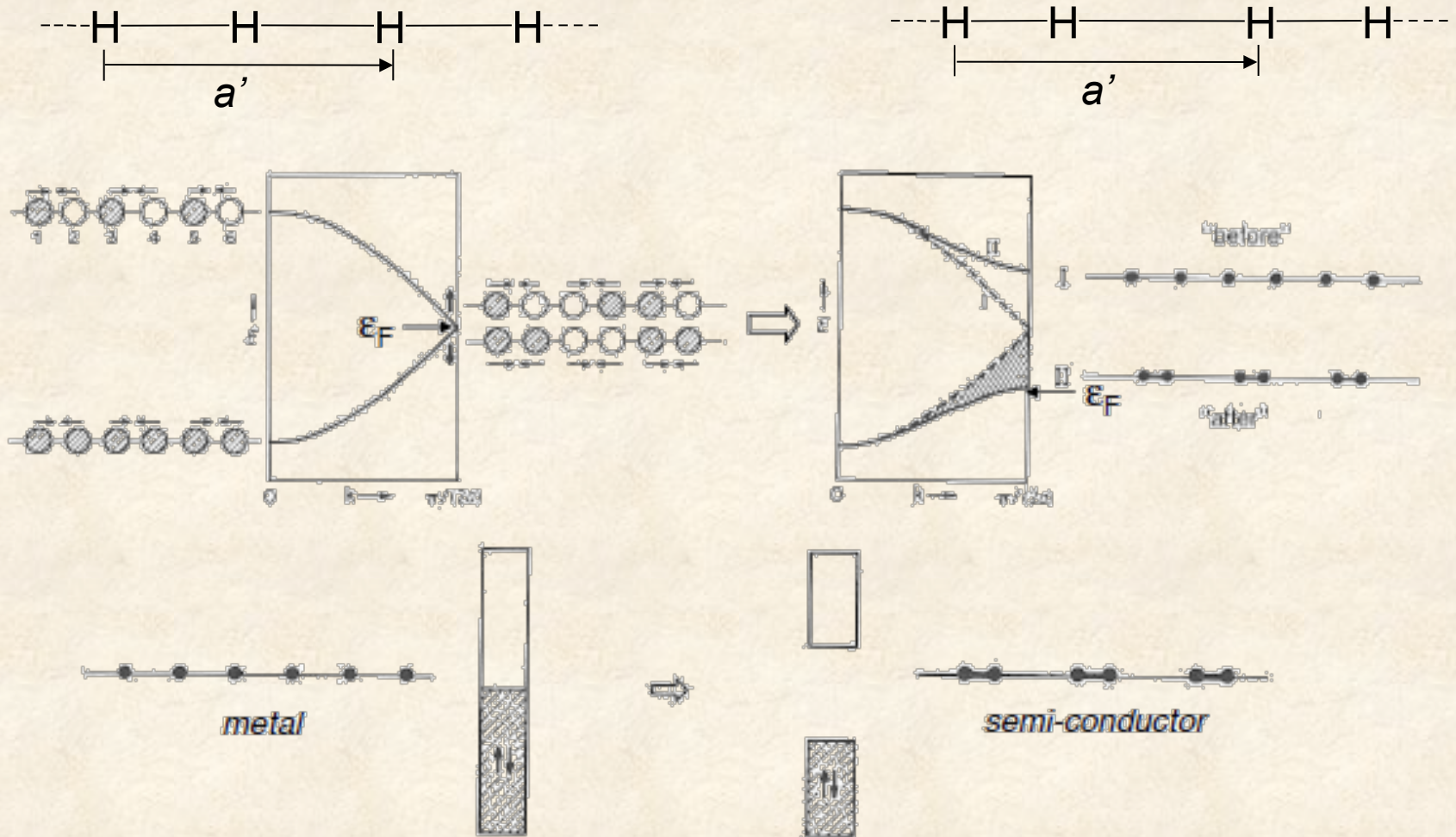
They interact strongly at $k = 0$ (S is maximum)

They do not interact by symmetry at $k = \pi/a'$ (S = 0)



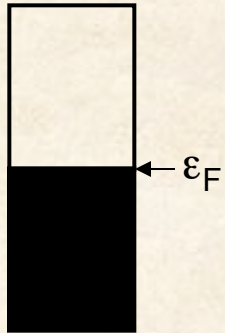
The Peierls Distortion

Peierls distortion (solid) \equiv Jahn-Teller distortion (molecule)

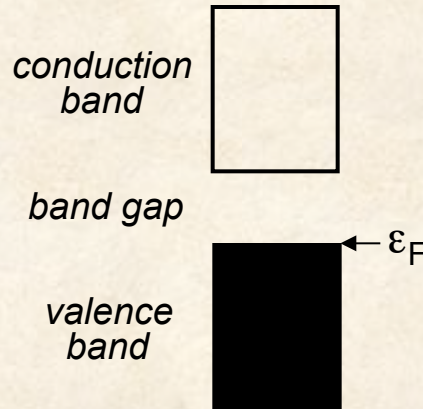


Metal, Semi-conductor or Insulator?

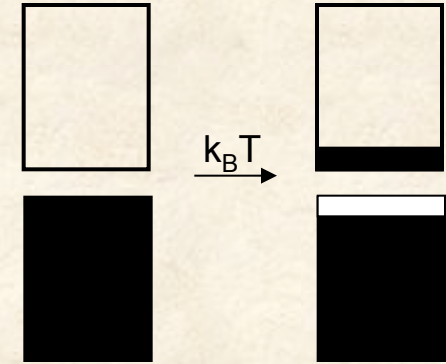
- At the "atomic" level



conductor (metal)

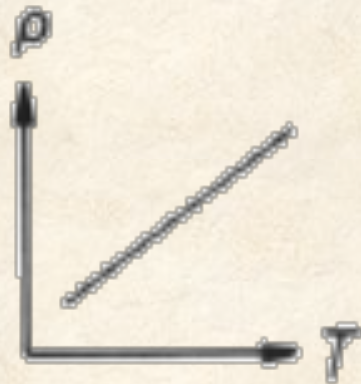


semi-conductor
or insulator

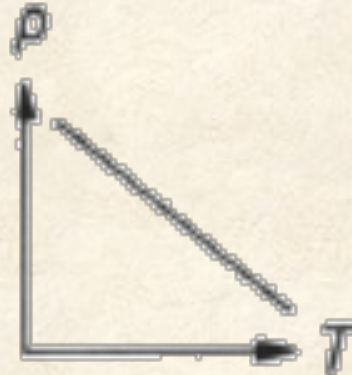


semi-conductor → metal

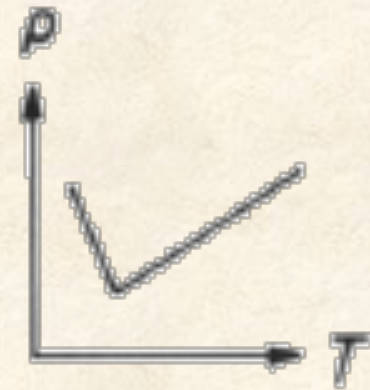
- At the macroscopic level (ρ = resistivity)



conductor (metal)
 $\rho \sim 10^{-6} \Omega \cdot \text{cm}$



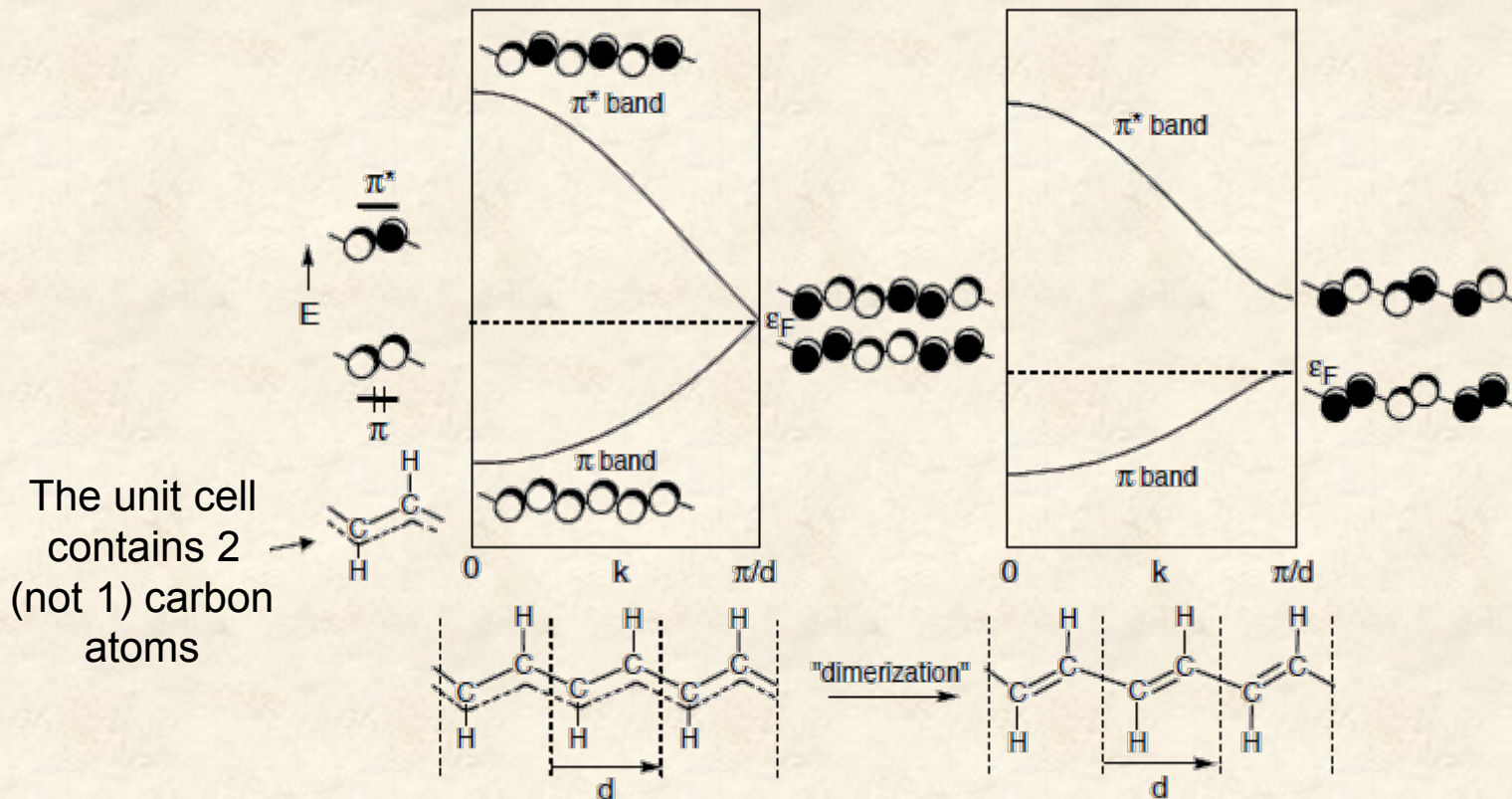
semi-conductor ($\rho \sim 10^5 \Omega \cdot \text{cm}$)
or insulator ($\rho \sim 10^{15} \Omega \cdot \text{cm}$)



semi-conductor → metal
($\rho \sim 10^5 \Omega \cdot \text{cm}$) ($\rho \sim 10^{-3} \Omega \cdot \text{cm}$)

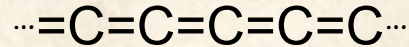
ρ (resistivity) inversely proportional to DOS $n(e)$ at ϵ_F

Exercise: Generate the π -type band structure of a regular zigzag polyacetylene chain and show that it is Peierls unstable

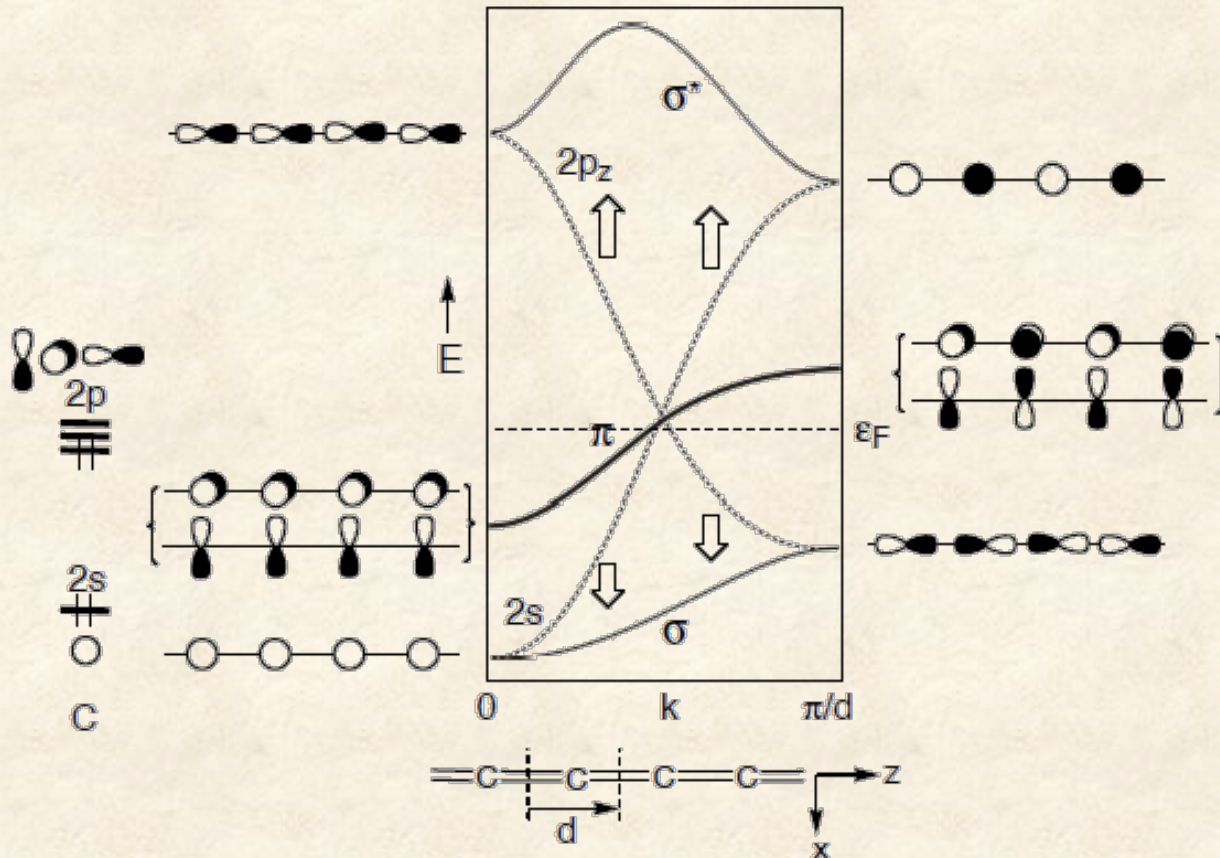


NB. The Peierls distortion is not associated with a doubling of the unit cell, but with the loss of a helicoidal axis.

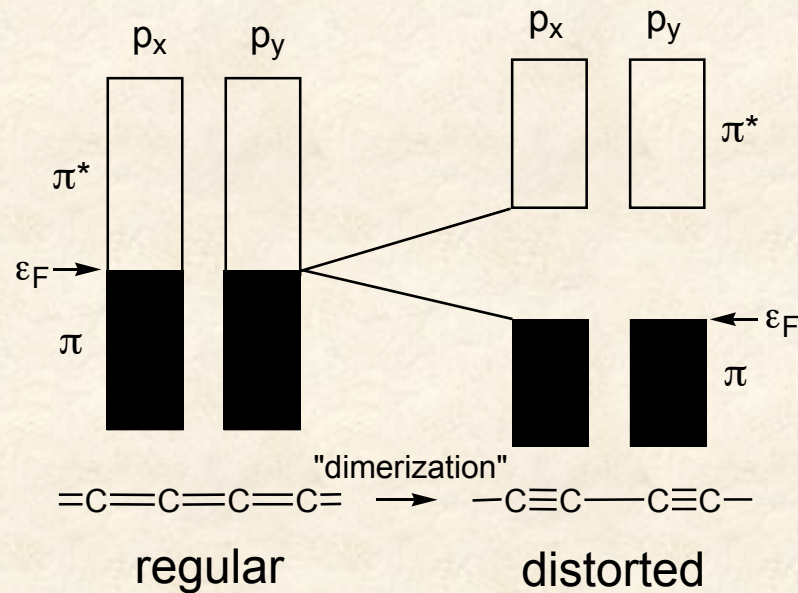
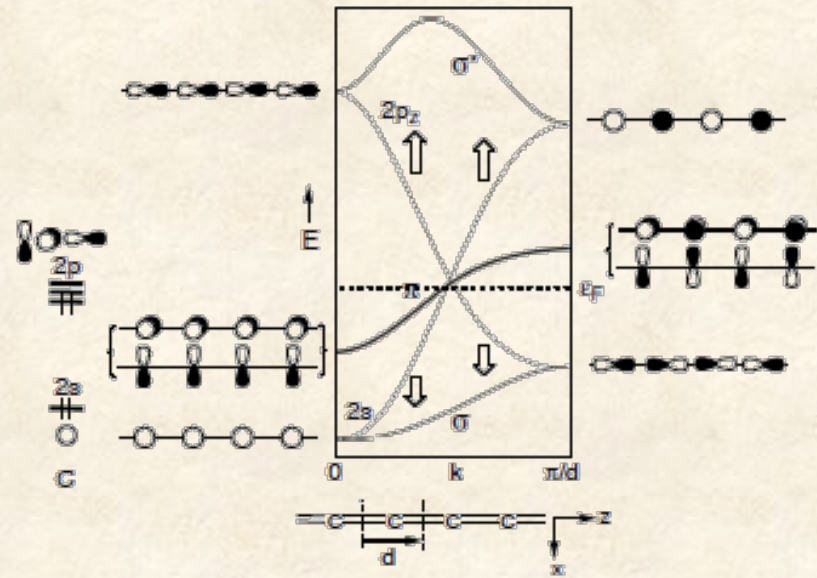
Exercise: Generate the band structure of a chain made of equidistant carbon atoms



- Each carbon AO generates a Bloch function.
- The σ -type 2s and $2p_z$ Bloch bands interact and strongly repel each other. The resulting σ and σ^* bands have no particular shapes and are rather flat.
- The π band is doubly degenerate.



A chain of equidistant carbon atoms... Peierls Instability...



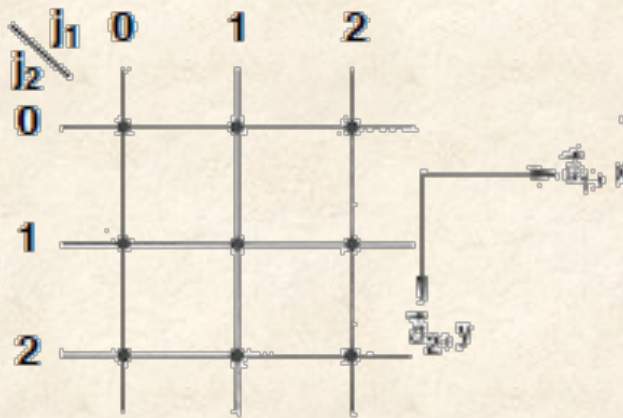
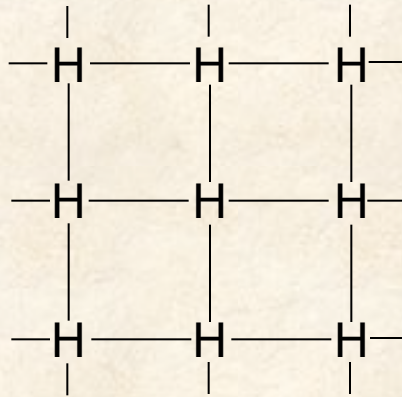
To summarize

- a) Identify the repeat unit (unit cell)
- b) Consider the content of the repeat unit as a molecular fragment
- c) Consider the MO diagram of this repeat unit (AOs if single atom). It contains the interactions between all the AOs inside one individual unit cell
- d) Generate the Bloch functions developed on each MO. A Bloch function developed on one particular χ MO of the unit cell describes the interactions between all the χ_j MOs of different unit cells
- e) Allow the Bloch orbitals to interact at each k -point if symmetry allows it to get the crystal orbitals (CO). The final band structure will arise

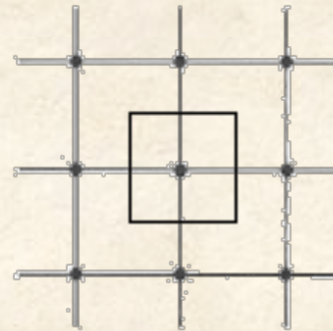
More Dimensions

Two Dimensions

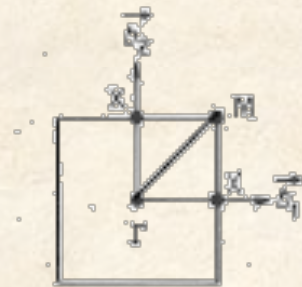
Square lattice of H atoms
(1 H per unit cell)



direct lattice



reciprocal lattice



Brillouin zone

Bloch orbitals: $\phi(k) \propto \sum_{j_1} \sum_{j_2} \exp(ikR_j) \chi_j$

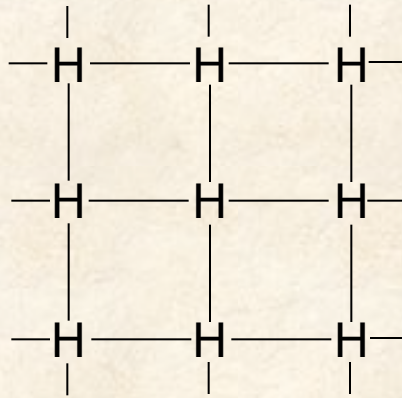
where $k = (k_x, k_y)$

$R_j = j_1 \mathbf{a}_1 + j_2 \mathbf{a}_2$ (j_1 and $j_2 = 0, \infty$)

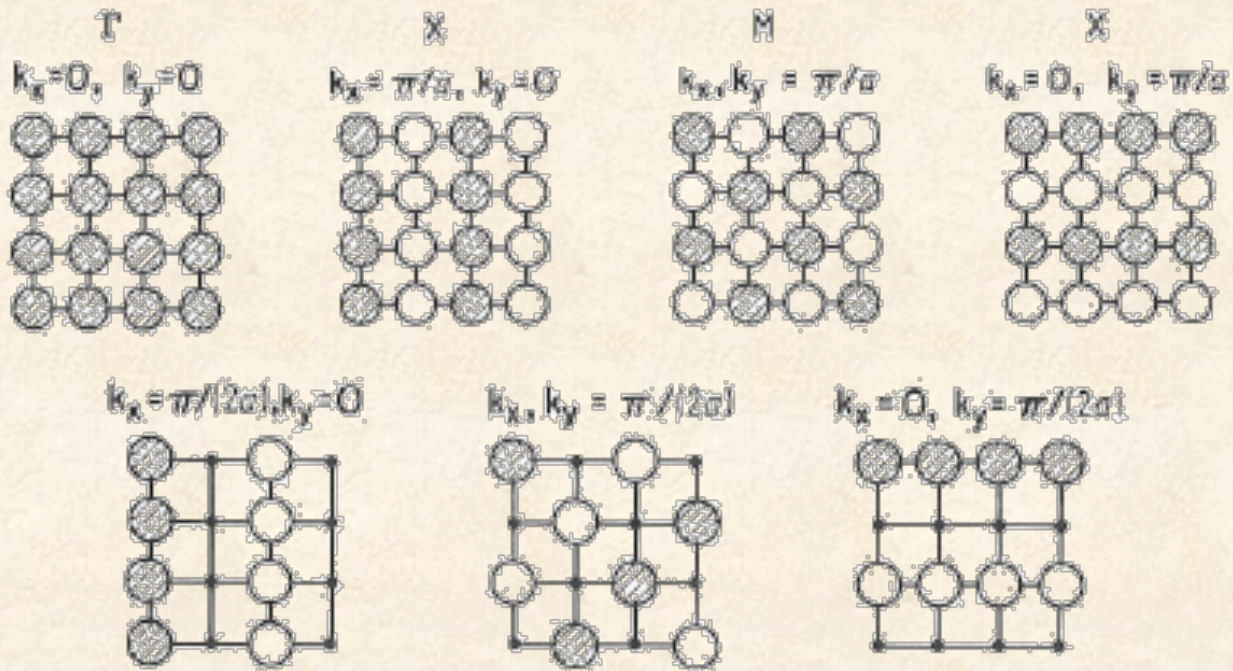
$\chi_j = 1s(\text{H})$

Two dimensions

Square lattice of H atoms
(1 H per unit cell)

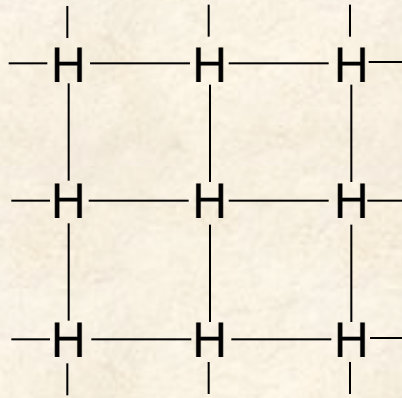


Bloch orbitals:



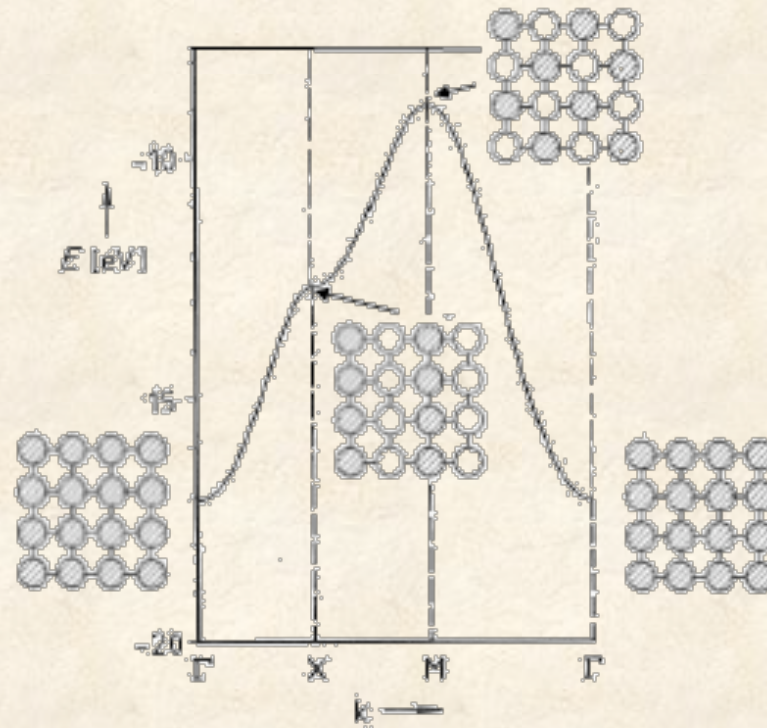
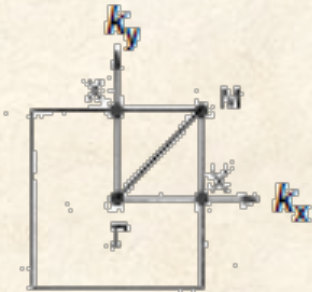
Two dimensions

Square lattice of H atoms
(1 H per unit cell)

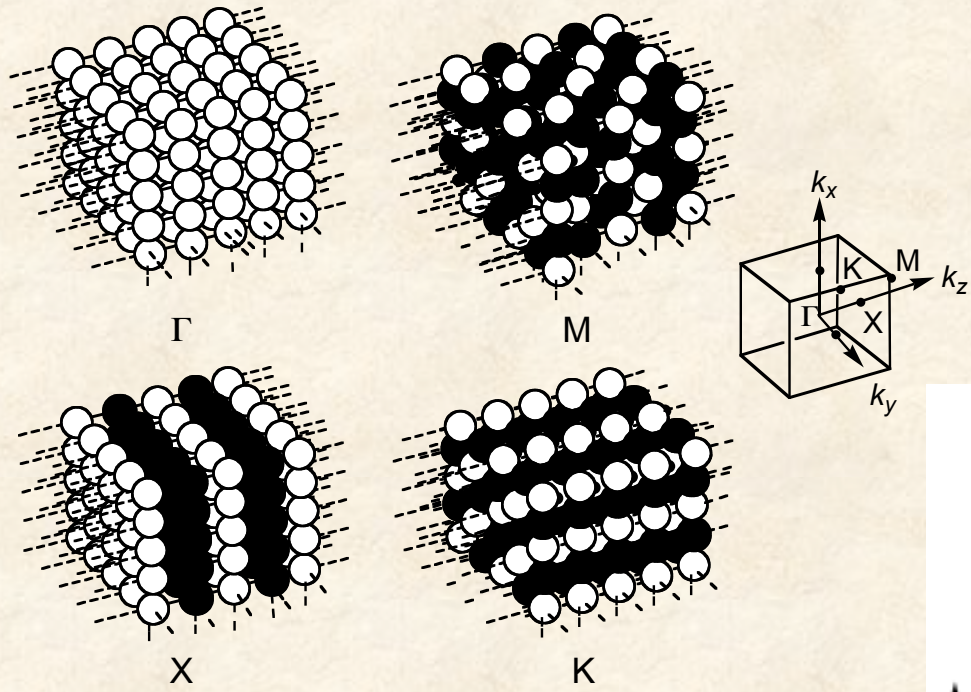


$$\phi(k) \longrightarrow E(k)$$

Band structure: $E(k) = f(k)$

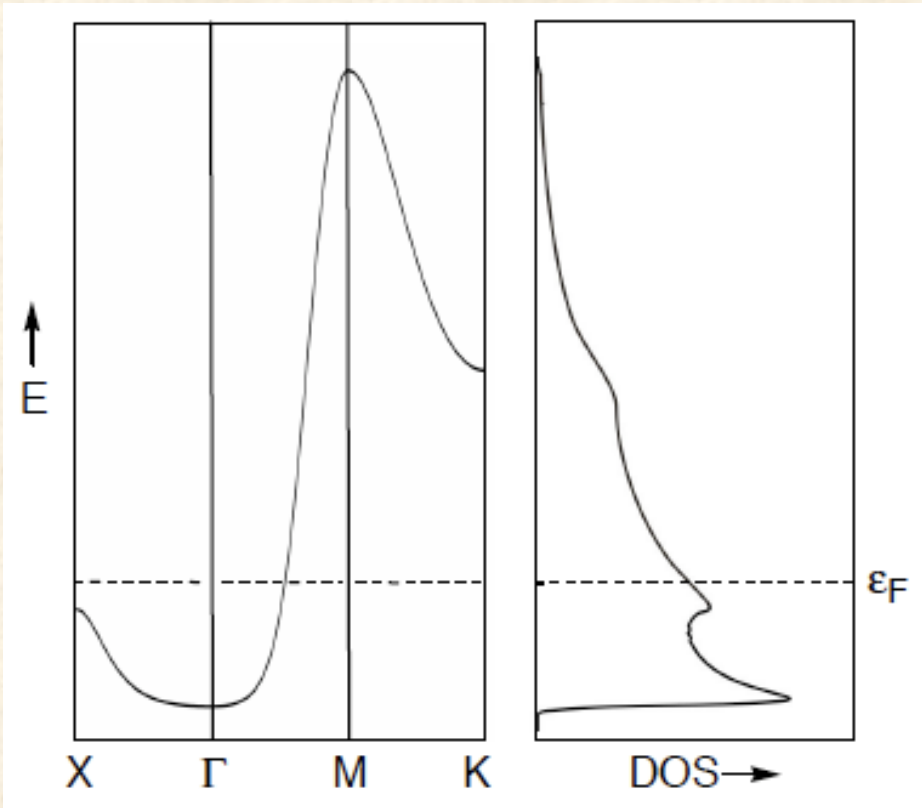


Three dimensions

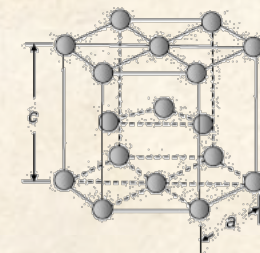
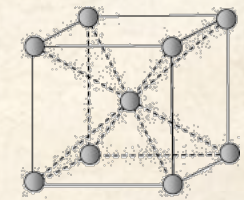
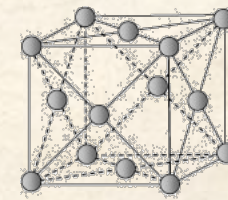
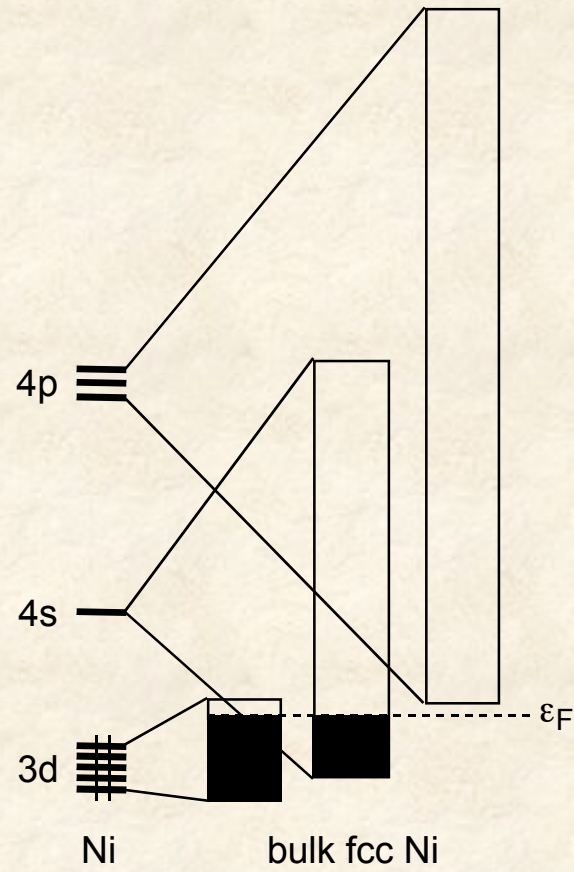
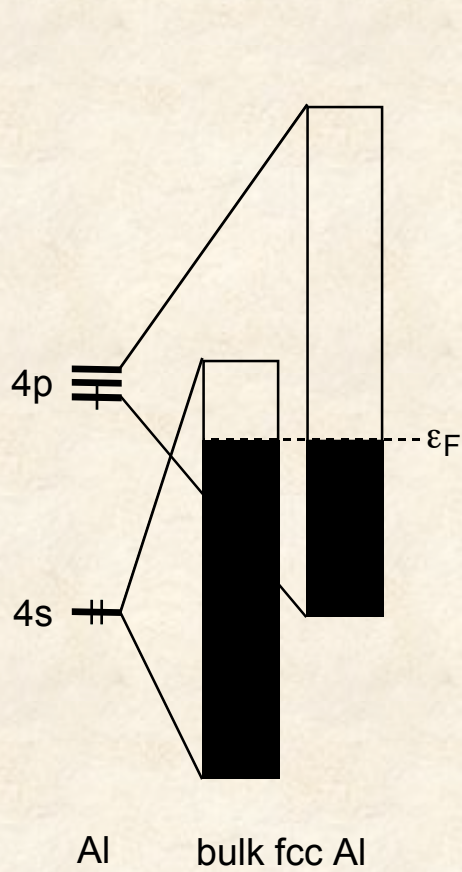


Simple cubic lattice of H atoms
(1 H per unit cell)

Bloch orbitals

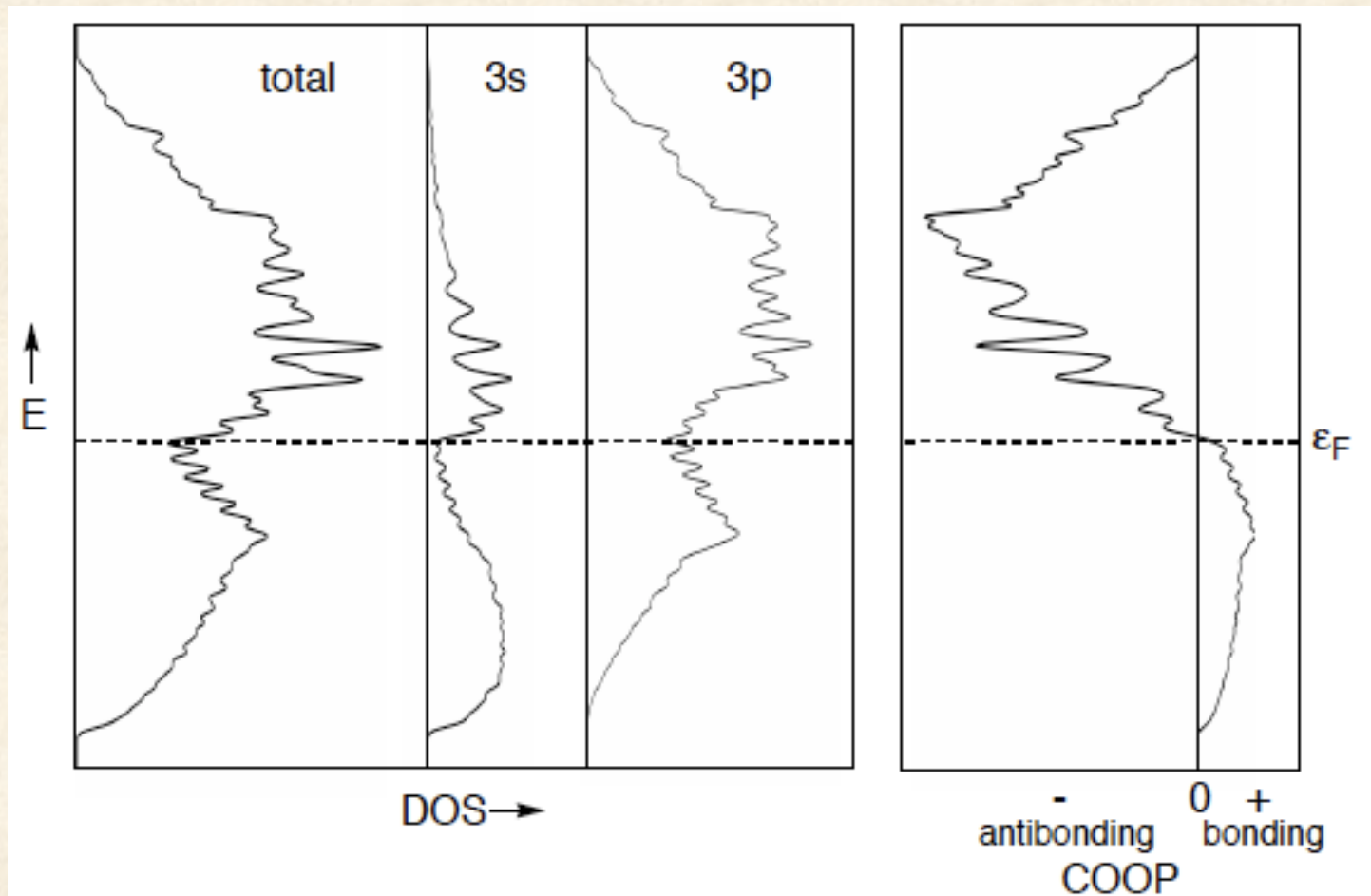


Three dimensions



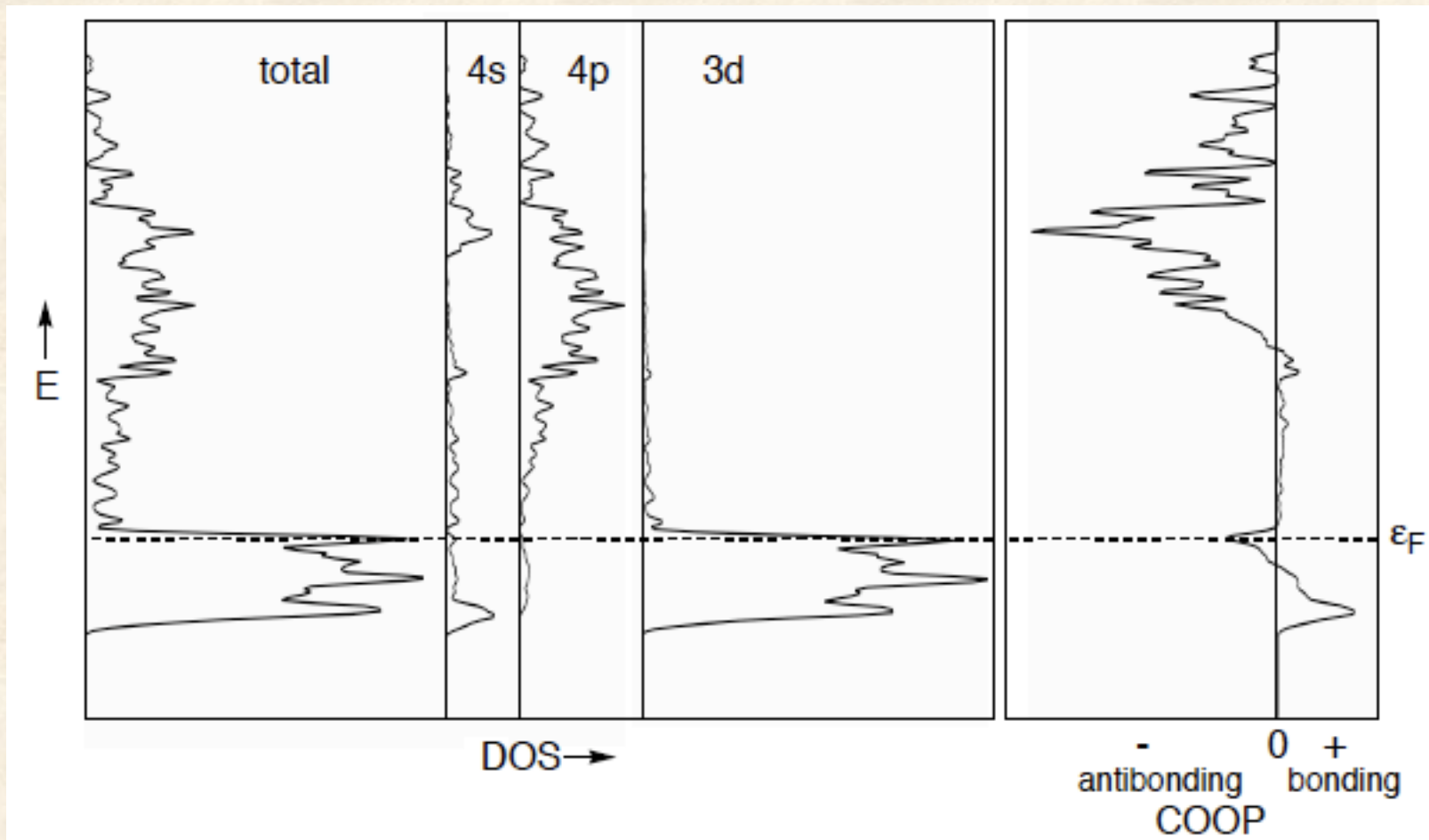
Three dimensions

Bulk fcc Al

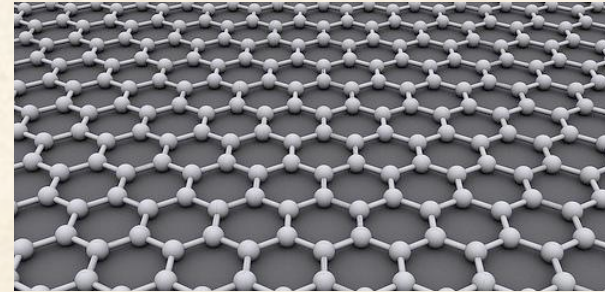
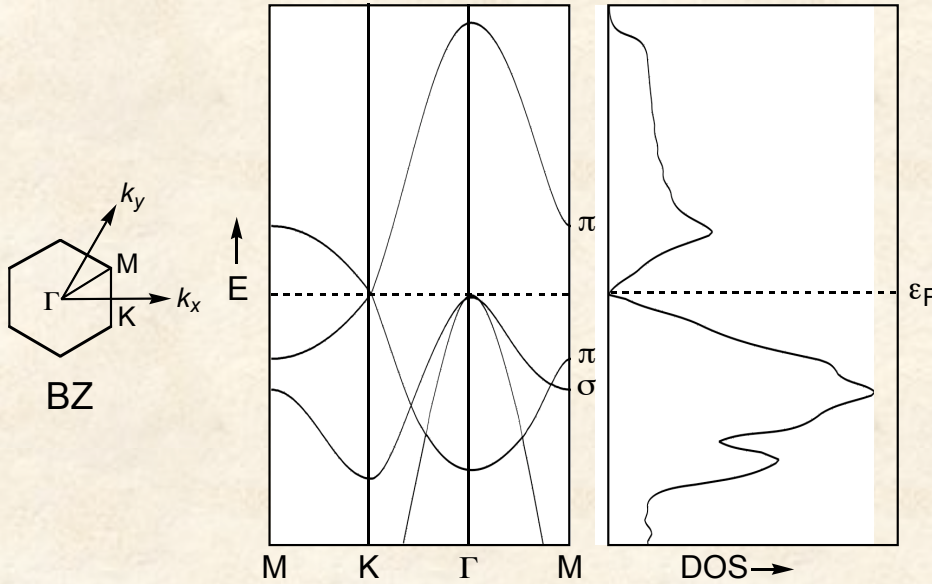


Three dimensions

Bulk fcc Ni

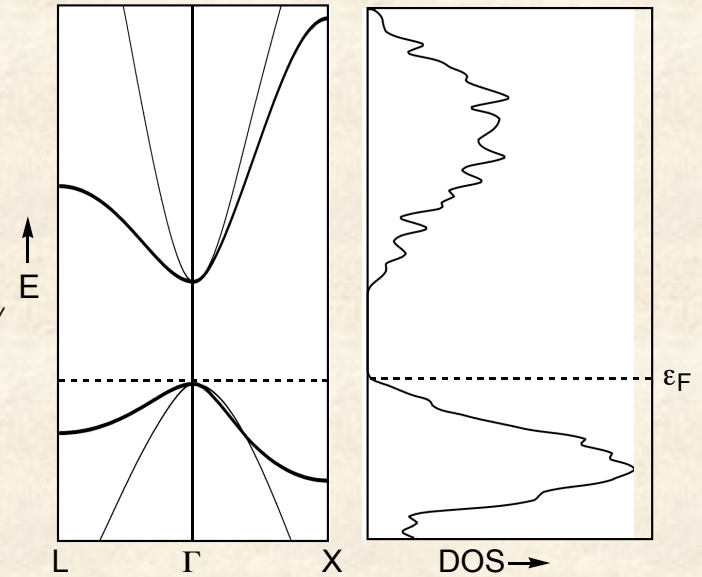
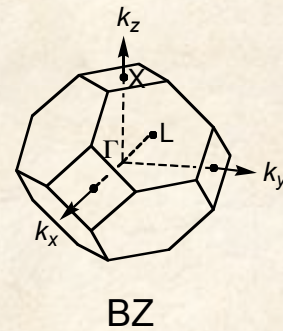
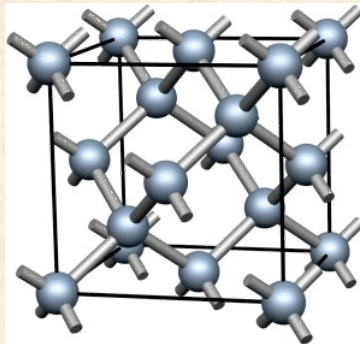


More examples: graphite and diamond



Diamond (3-D)

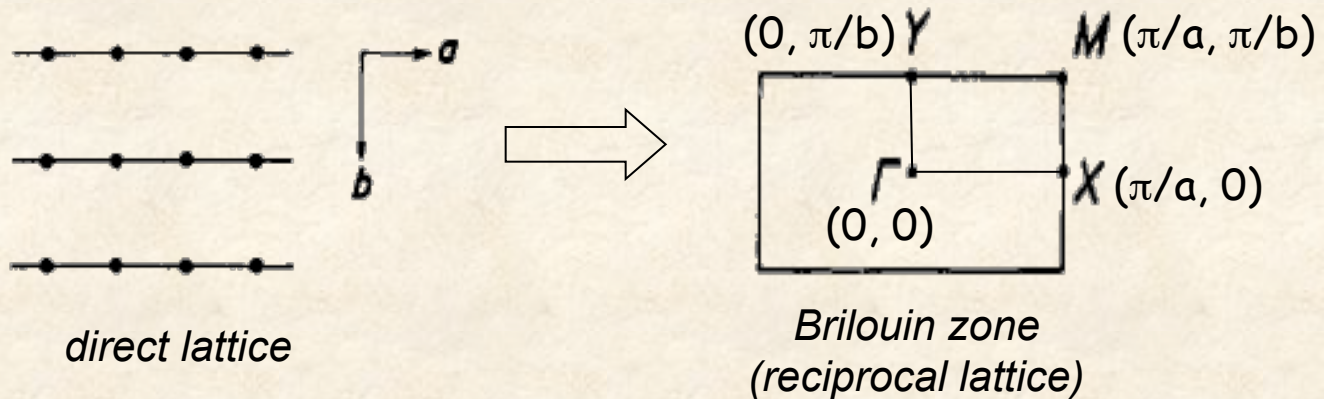
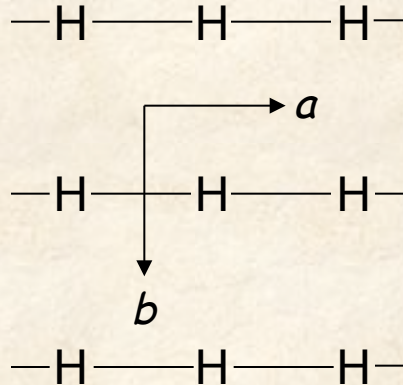
Large forbidden energy gap above ϵ_F
Insulator



Partially Filled Bands: The Fermi Surface

Partially Filled Bands: The Fermi Surface

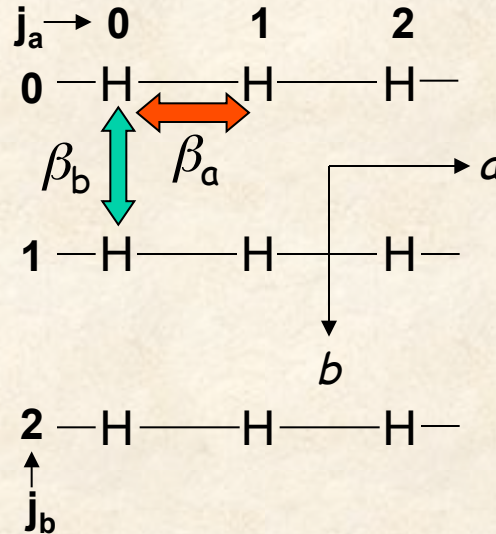
Rectangular lattice of H atoms
(1 H per unit cell)



Construction of BOs, $\phi(k) \implies$ Band structure, $E(k) = f(k)$

Partially Filled Bands: The Fermi Surface

Rectangular lattice of H atoms
(1 H per unit cell)

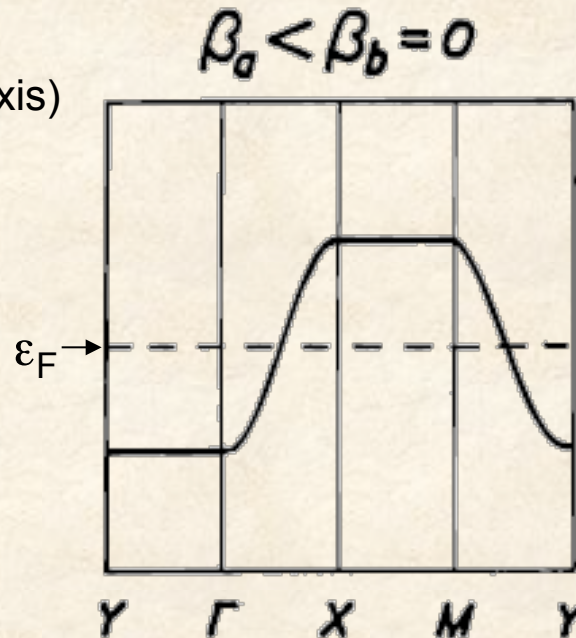


β_a and β_b : resonance integrals
(transfer integrals)

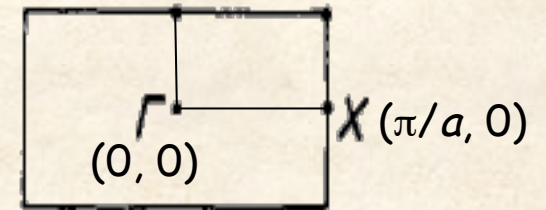
$$\beta_a = \langle \chi_{j_a} | H^{\text{eff}} | \chi_{j_a+1} \rangle$$

$$\beta_b = \langle \chi_{j_b} | H^{\text{eff}} | \chi_{j_b+1} \rangle$$

Band structure, $E(k) = f(k)$
(if no interaction along b axis)



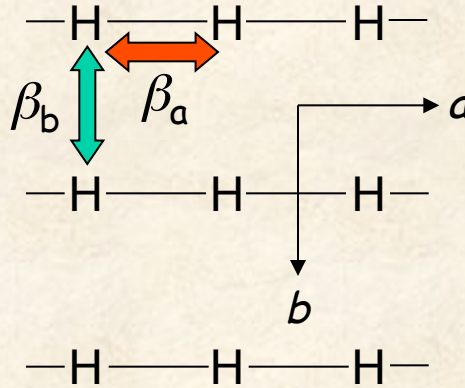
$(0, \pi/b) Y$ $M (\pi/a, \pi/b)$



Brilouin zone

Partially Filled Bands: The Fermi Surface

Rectangular lattice of H atoms
(1 H per unit cell)



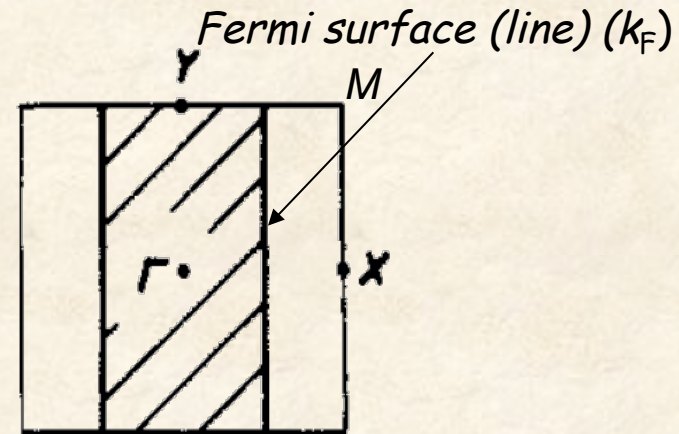
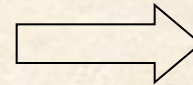
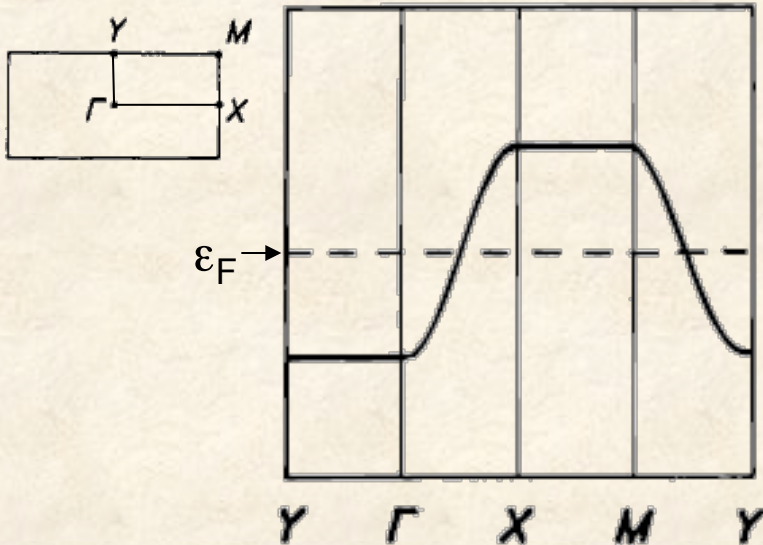
β_a and β_b : resonance integrals
(transfer integrals)

$$\beta_a = \langle \chi_{ja} | H^{\text{eff}} | \chi_{ja+1} \rangle$$

$$\beta_b = \langle \chi_{jb} | H^{\text{eff}} | \chi_{jb+1} \rangle$$

Partially filled bands \Rightarrow Fermi surface

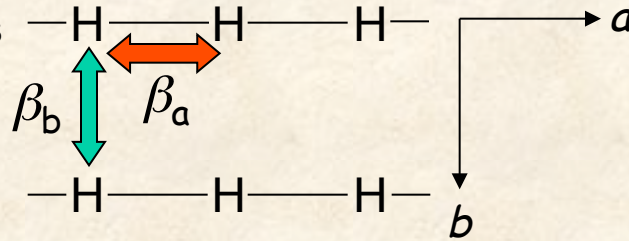
$$\beta_a < \beta_b = 0$$



Fermi surface: boundary surface separating the “occupied” and “unoccupied” wave vectors: a point (1D), a line (2D) or a surface (3D)

Partially Filled Bands: The Fermi Surface

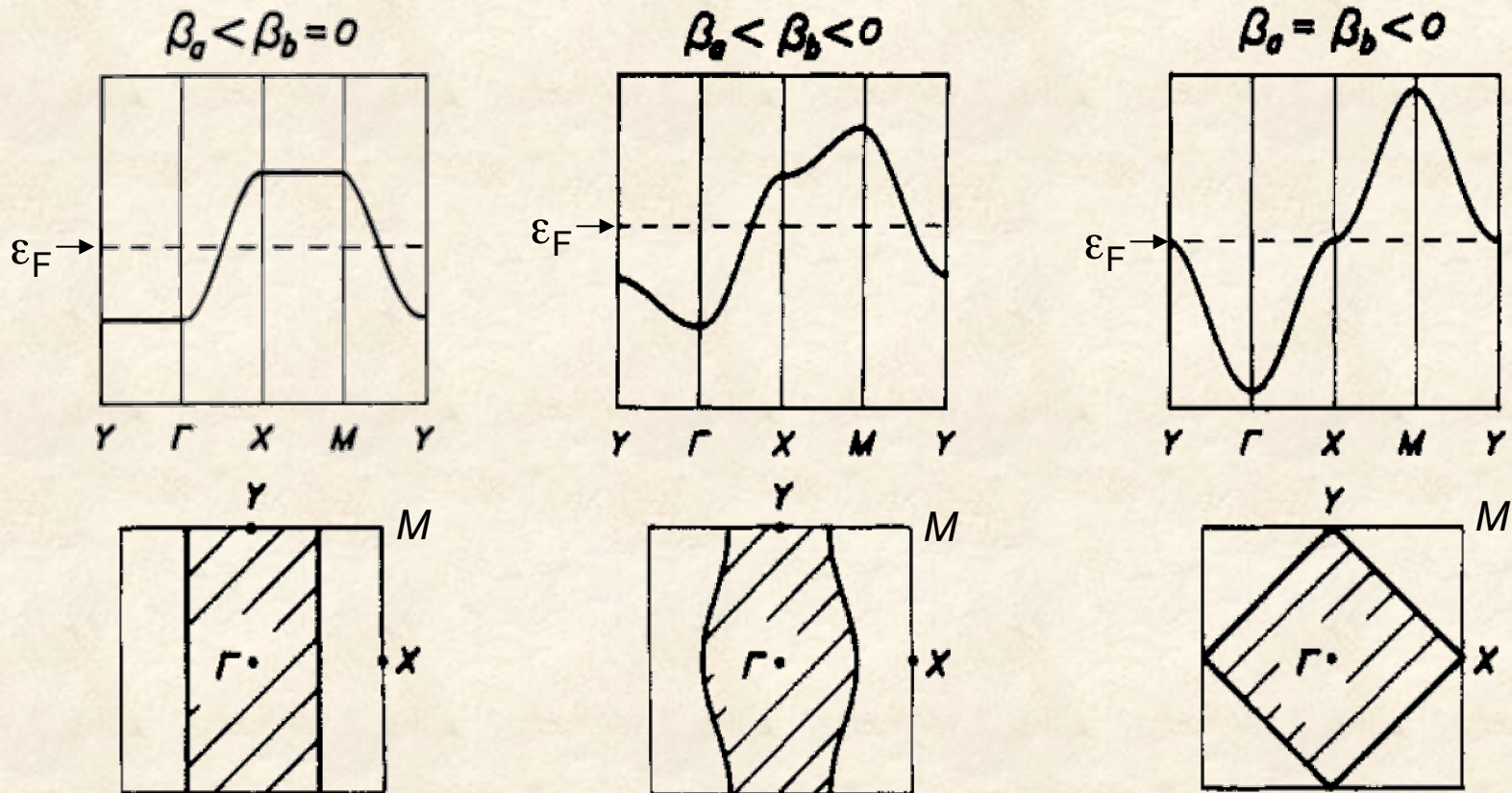
Rectangular lattice of H atoms
(1 H per unit cell)



$$\beta_a = \langle \chi_{ja} | H^{\text{eff}} | \chi_{ja+1} \rangle$$

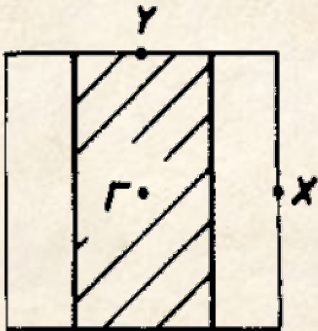
$$\beta_b = \langle \chi_{jb} | H^{\text{eff}} | \chi_{jb+1} \rangle$$

Band structures and Fermi surfaces

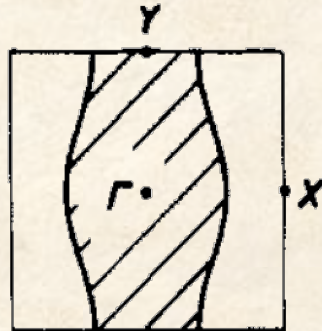


Partially Filled Bands: The Fermi Surface

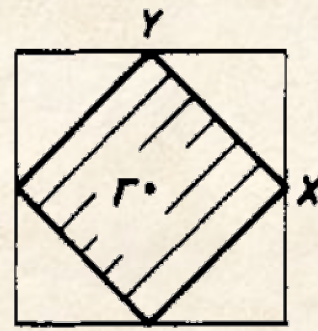
Fermi surfaces explain the dimensionality of metallic properties



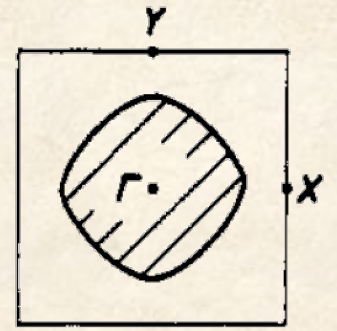
1D metal
(along a axis)



pseudo-1D metal
(along a axis)



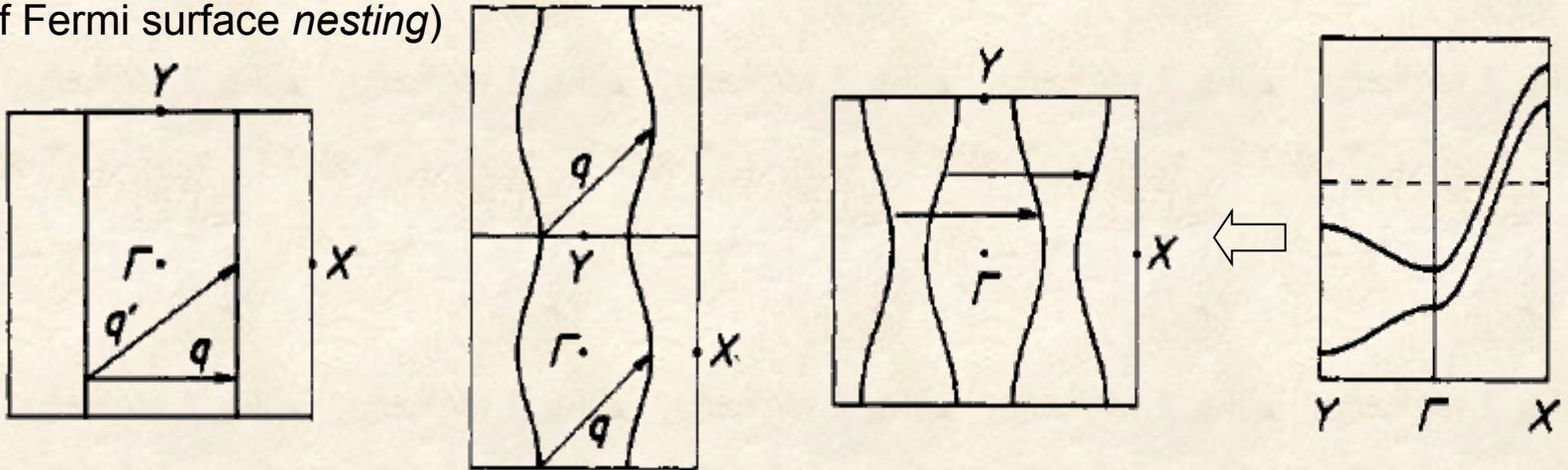
2D (anisotropic) metal
(along $a+b$ and $a-b$ axis)



2D (isotropic) metal
(close loop)

Partially Filled Bands: The Fermi Surface

Fermi surfaces explain distortions in solids due to electronic instabilities
(concept of Fermi surface *nesting*)

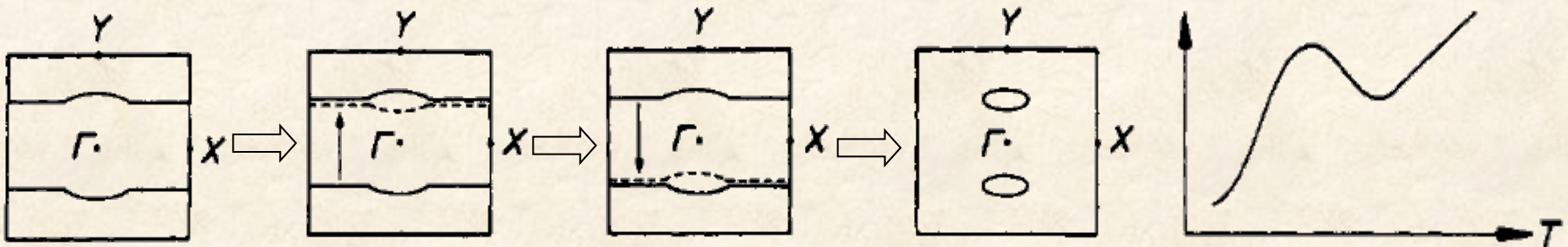


A Fermi surface is nested by a vector q when a section can be moved by such a vector to be superimposed on another section

A metallic system with a *nested* Fermi surface is subject to a metal-insulator phase transition
(interaction of occupied and vacant crystal orbitals of same symmetry of $E = E(k_F)$ related by q)

If complete \Rightarrow destruction of the Fermi surface

If incomplete \Rightarrow Fermi surface with hole and/or electron pockets



Computational Aspects

Computational Aspects

- Quantum chemical methods

There are different ways to solve the Schrödinger equation
for periodic systems



Crystal orbitals $\Psi_n(k)$

SCHRÖDINGER EQUATION

$$H\Psi = E\Psi$$

$\Psi(r)$

Born-Oppenheimer approximation

$\rho(r)$

orbital approximation

Hohenberg-Kohn theorems

HARTREE-FOCK

LCAO approximation

DENSITY FUNCTIONAL THEORY

LCAO approximation

ROOTHAN

HÜCKEL,
EHT,
EH-TB⁴

Kohn-Sham (KS) development

KS-LDA(LSD) METHODS

non-LCAO

Mixed

LCAO

PW-PP¹, APW
OPW, LMTO²,
MS-X α ...

FP-LAPW³

LCGTO-LSD,
FPLO, KKR...

AB INITIO SCF

approximation on Hamiltonian

treatment of correlation

SEMI-EMPIRICAL MODELS:
AM1, PM3, CNDO,
INDO, MNDO...

treatment of exchange-correlation

KS NON-LDA (GGA) METHODS:
BP86, B3LYP, PBE...

treatment of excited states

MCSCF, CI, CC
MPn/MBPT...

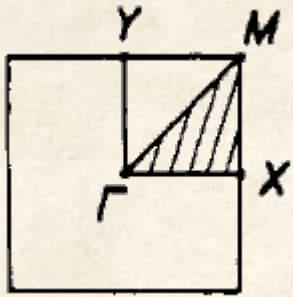
TD-DFT

¹VASP, ²LMTO-ASA, ³WIEN-2k, ⁴Yahmop

Computational Aspects

Crystal orbitals $\Psi_n(k) \iff e_n(k)$
for specific k wave vectors along symmetry lines
of the irreducible part of Brillouin zone

- Band dispersion relation, $E(k) = f(k)$



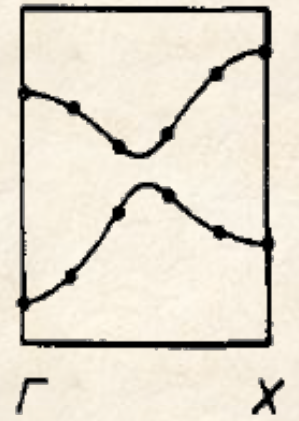
Brillouin zone



one band



*two bands with
different symmetry*

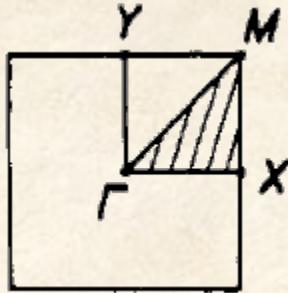


*two bands with
identical symmetry*

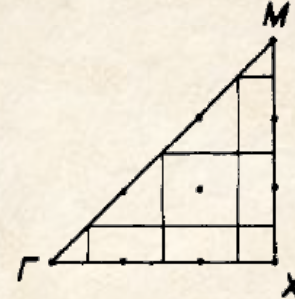
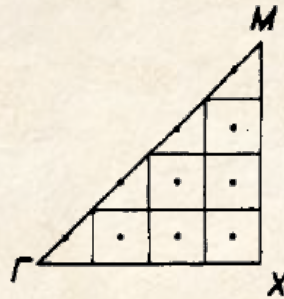
Computational Aspects

M crystal orbitals $\Psi_n(k) \implies e_n(k)$

for L k wave vectors “inside” the irreducible part of Brillouin zone (sampling)

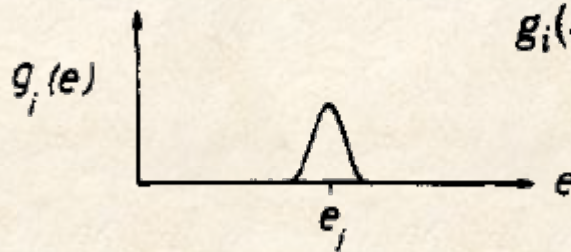
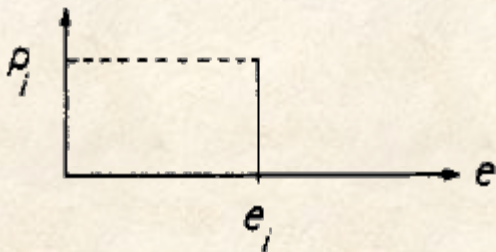


Brillouin zone



two sets of 10 k “points” of weight w_i

- Density of states, Fermi level, and electronic energy



$$g_i(e) = \frac{1}{\sqrt{\pi}\delta} p_i \exp \left[-\left(\frac{e - e_i}{\delta} \right)^2 \right]$$

$$\int_{-\infty}^{\infty} g_i(e) de = p_i$$

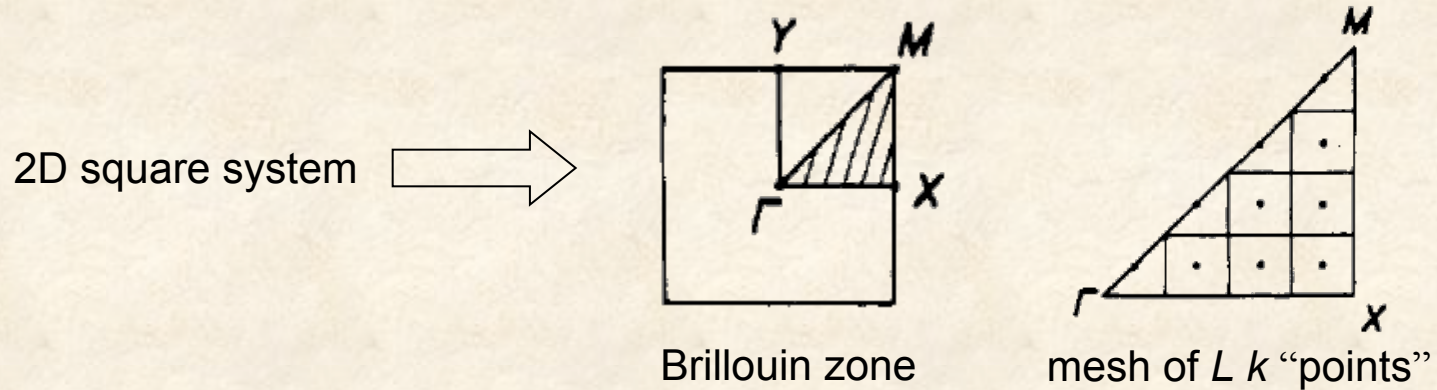
Total DOS: $n(e) = \sum_{i=1}^{M \times L} 2g_i(e)$ Fermi level: $\sum_{i=1}^{\max} 2p_i = N_{\text{tot}}$ (e-/unit cell) $\implies e_{\text{max}} = e_F$

Total electronic energy of the system: $E_{\text{elec}} = \int_{-\infty}^{e_f} n(e)e de$

Computational Aspects

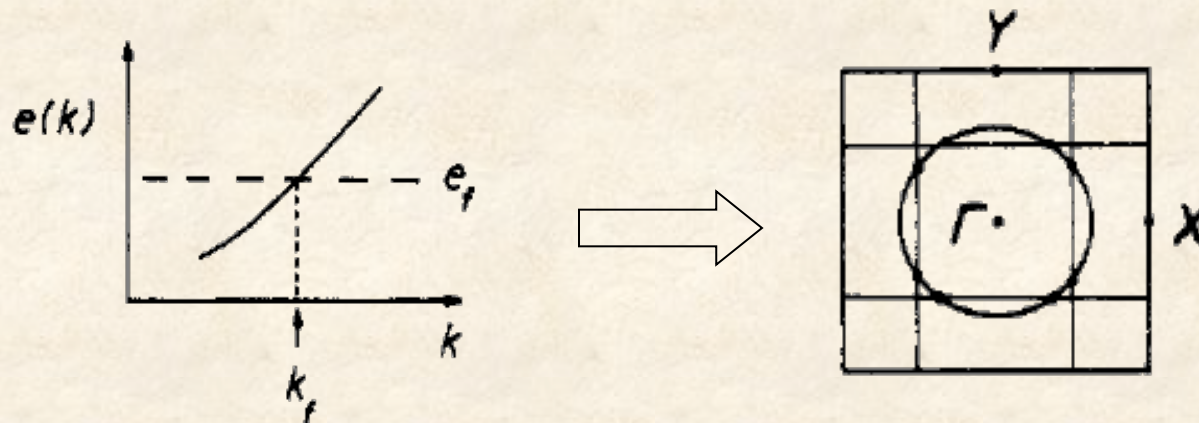
$$M \text{ crystal orbitals } \Psi_n(k) \implies e_n(k)$$

for L k wave vectors “inside” the irreducible part of Brillouin zone



- Fermi surface

$$e_n(k_i) \implies e_n(k_F) = e_F \text{ (constant energy surface)}$$



The Theoretical Machinery at Work...

Application to the Textbook Example of TTF-TCNQ, the First Organic Molecular Metal

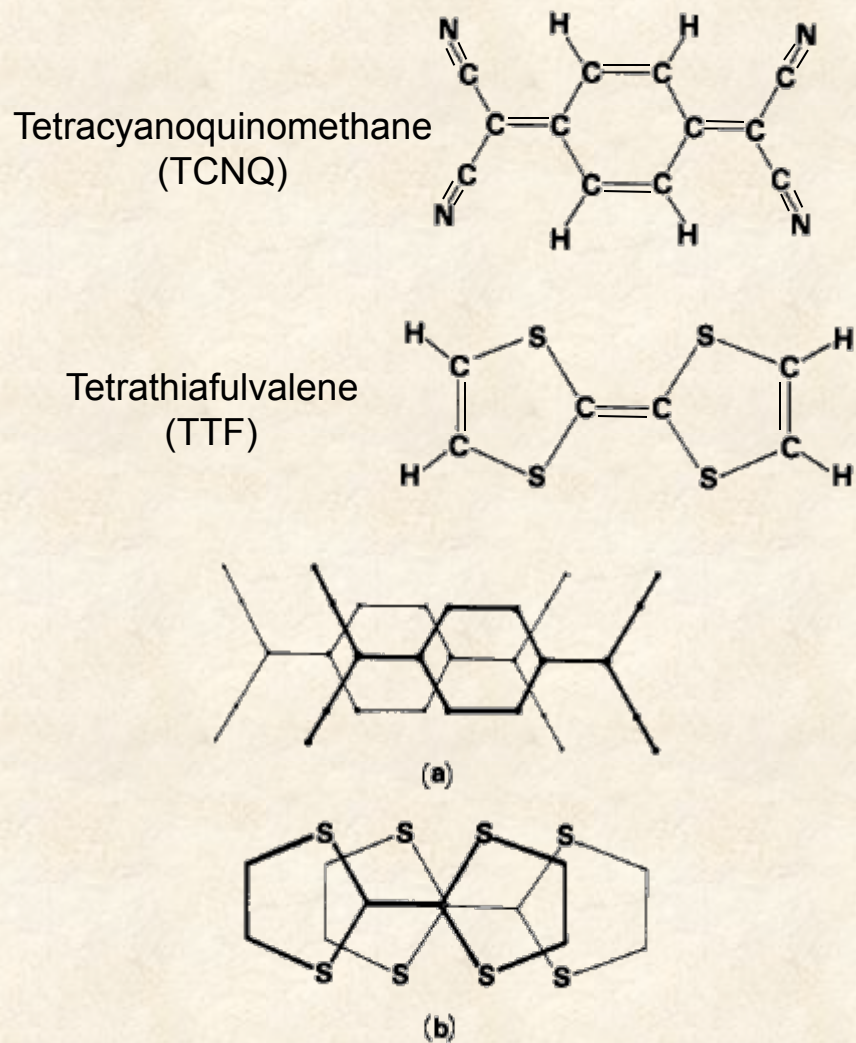


Figure 2. Arrangements found for neighboring molecules in stacks of (a) TCNQ, (b) TTF in TTF-TCNQ crystals.

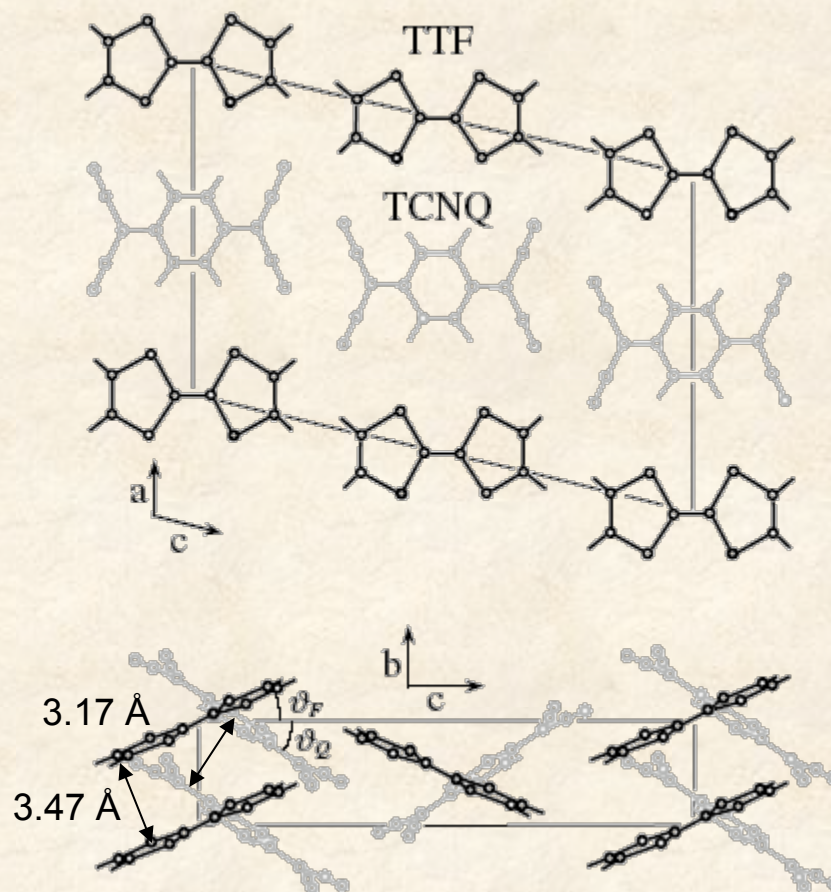


FIG. 1. Crystal structure of TTF-TCNQ.

The Textbook Example of TTF-TCNQ

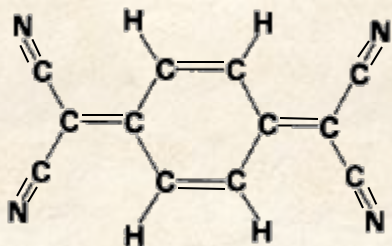
TTF-TCNQ: A truly organic molecular metal

$\sigma \sim 5 \cdot 10^3 \Omega^{-1} \cdot \text{cm}^{-1}$ at 80 K (-210 °C) (10^6 for Cu)

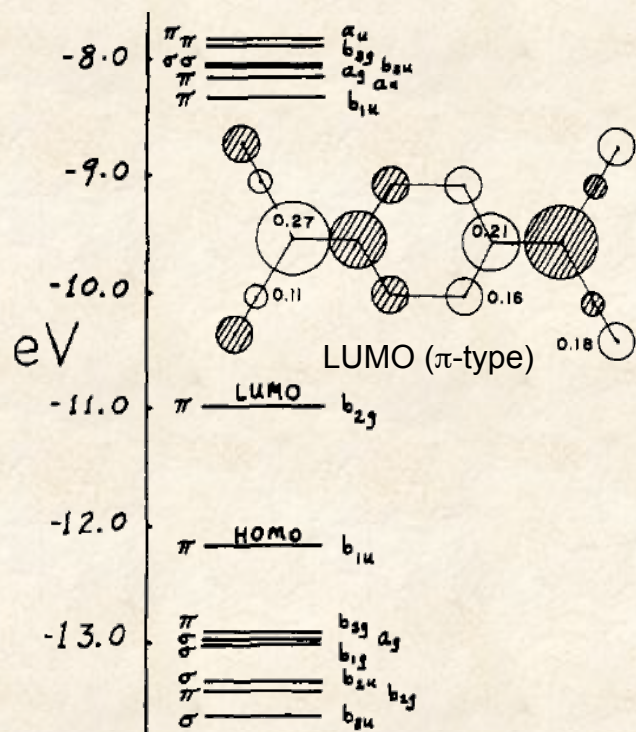
What's special for TTF-TCNQ?

The Textbook Example of TTF-TCNQ

TCNQ: A good acceptor

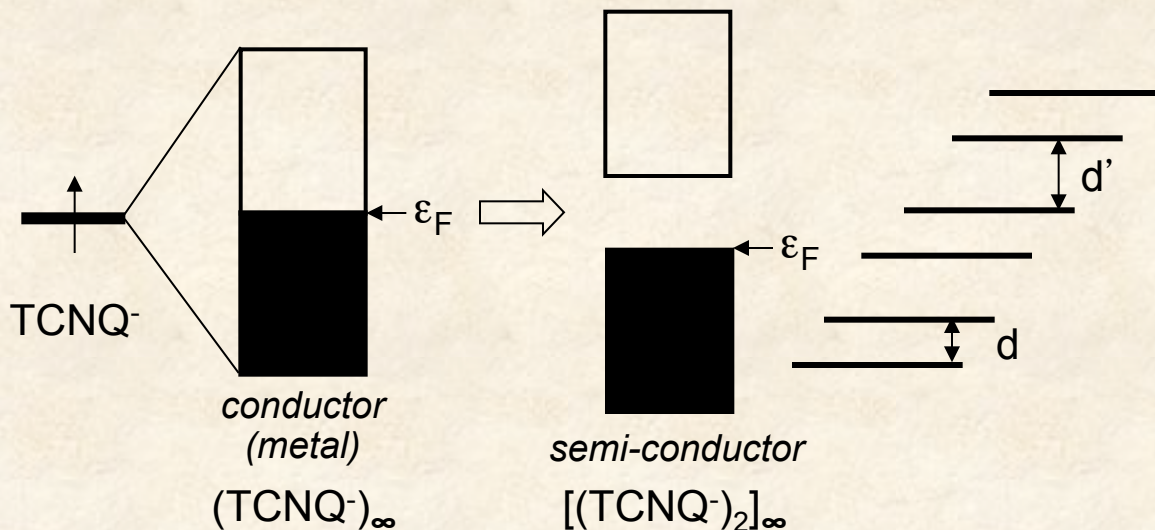


Tetracyanoquinomethane
(TCNQ) - $16 \pi e^-$

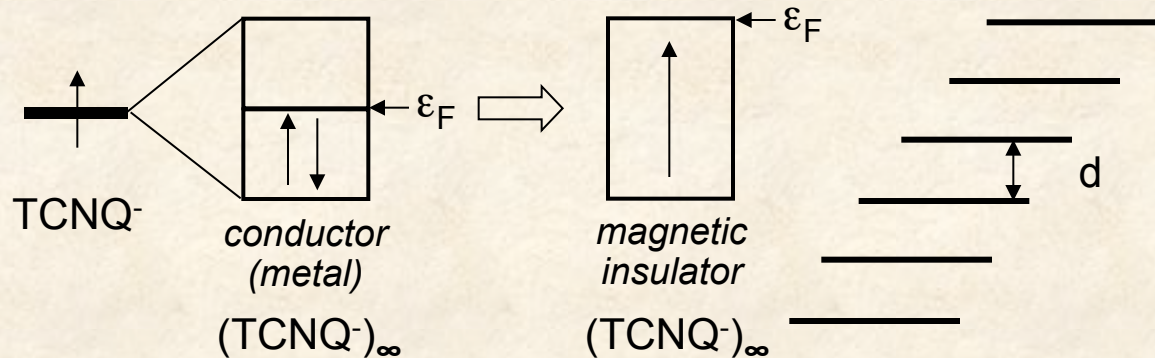


Some EHMO energy levels and MO symmetries for TCNQ.

weakly antibonding LUMO (π -type) \Rightarrow acceptor character \Rightarrow TCNQ⁻
(structural change upon reduction)

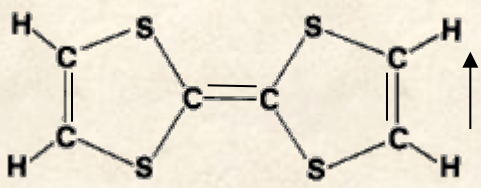


or alternatively, if U (e^-e^- repulsion) $>$ W (bandwidth)

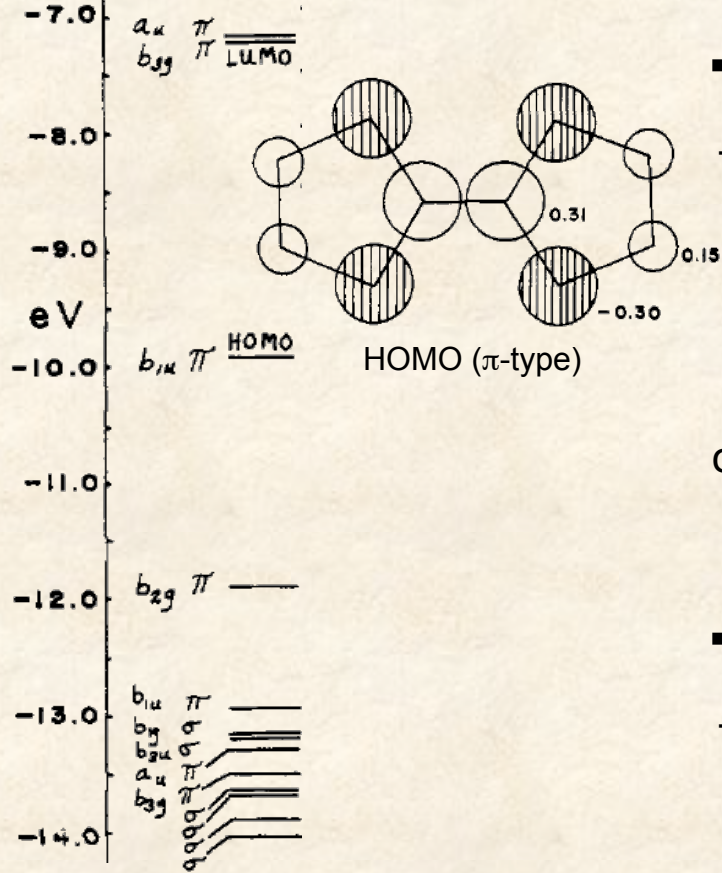


The Textbook Example of TTF-TCNQ

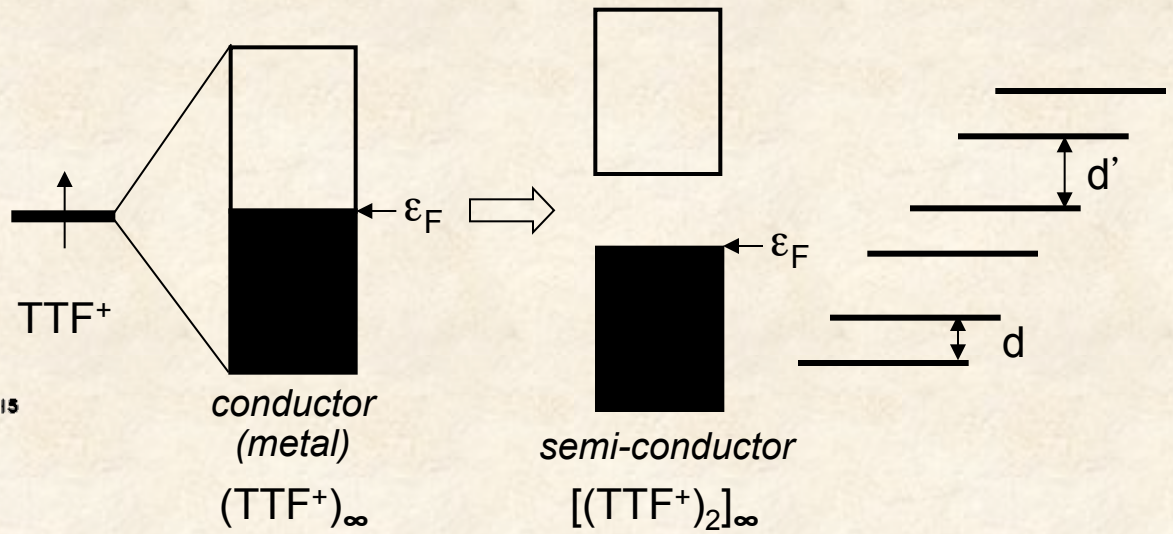
TTF: A good donor



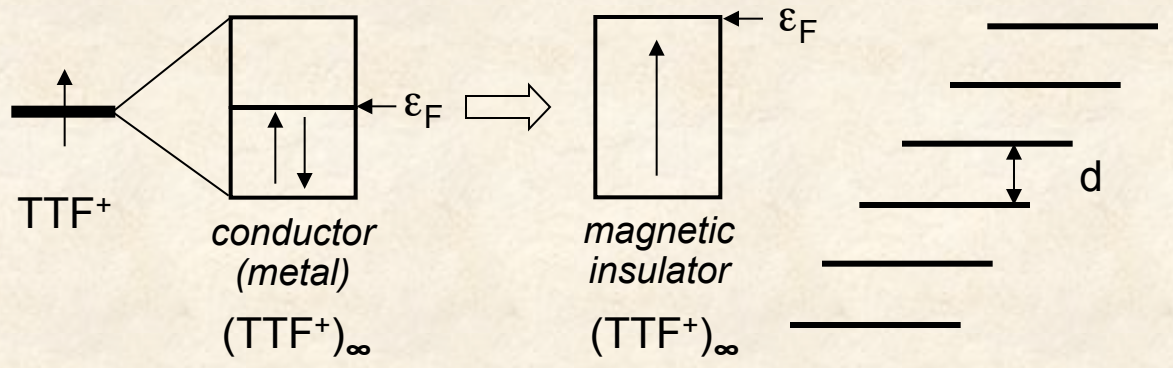
Tetrathiafulvalene (TTF) - $14 \pi e^-$



weakly bonding HOMO (π -type) \implies donor character \implies TTF⁺
 (structural change upon oxidation)



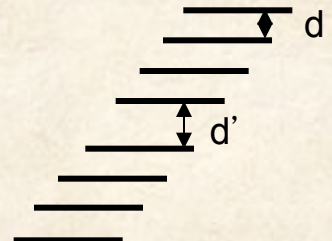
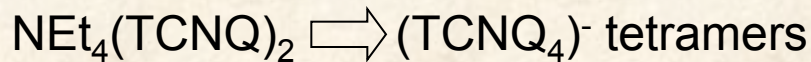
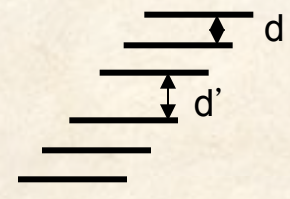
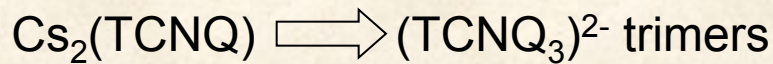
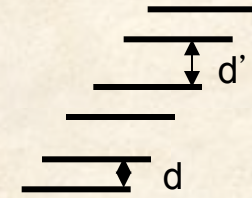
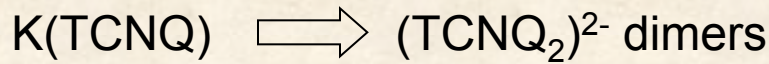
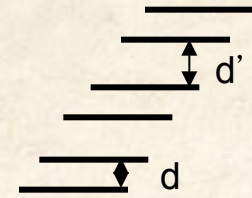
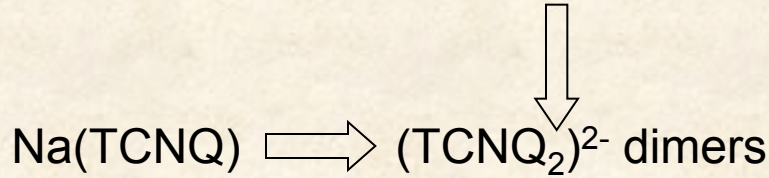
or alternatively, if U (e^-e^- repulsion) $>$ W (bandwidth)



Some EHMO energy levels and MO symmetries for TTF.

The Textbook Example of TTF-TCNQ

TCNQ salts are generally semi-conductors



TTF salts as well...

The Electronic Structure of TTF-TCNQ

What's special for TTF-TCNQ?

DFT band structure (two TCNQ and two TTF molecules per unit cell):

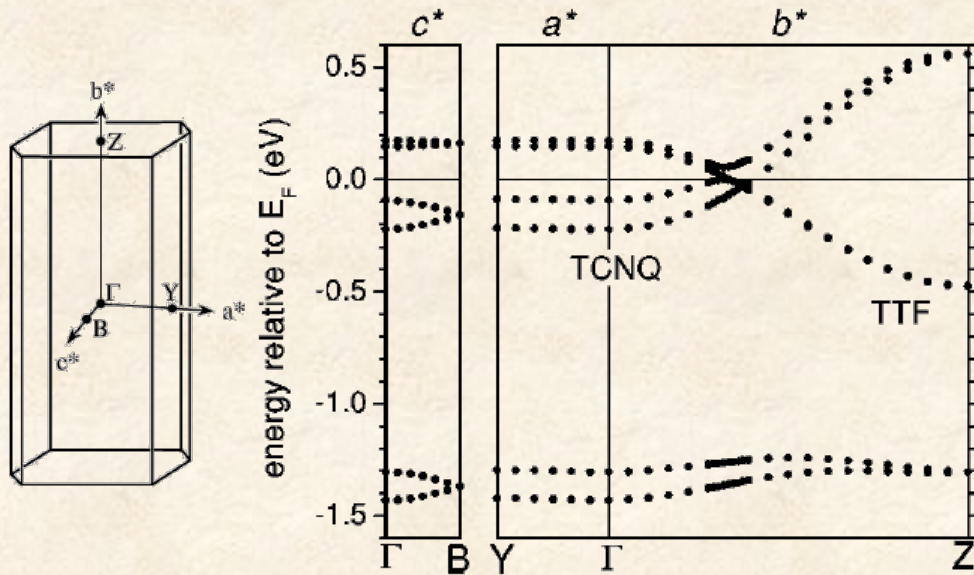


FIG. 2. DFT band structure near the Fermi level along the three major high-symmetry lines of the Brillouin zone of TTF-TCNQ.

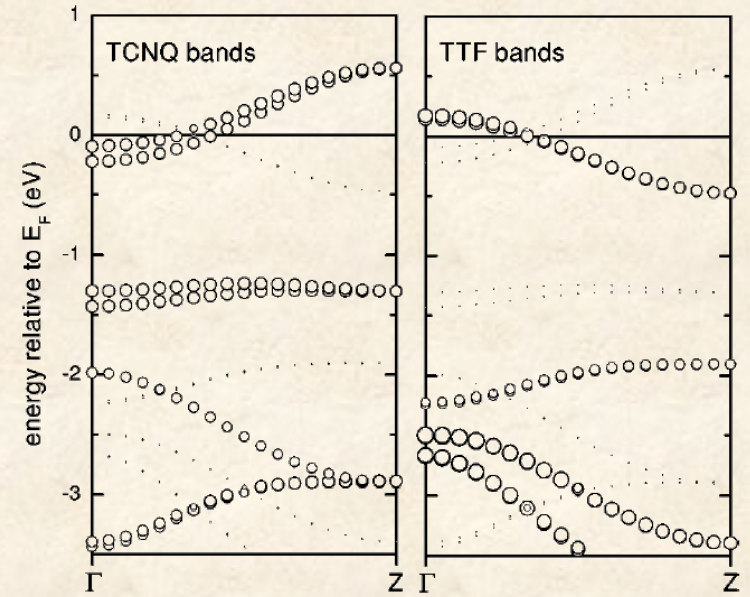


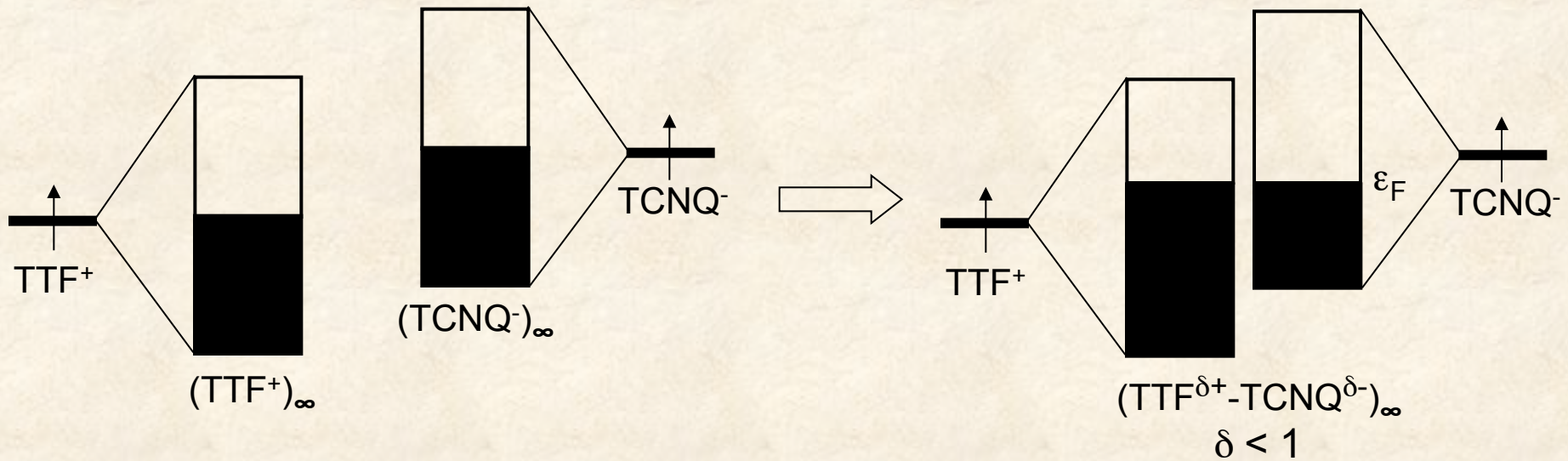
FIG. 3. Theoretical band dispersions along Γ Z showing the molecular origin of the bands. The size of the symbols represents the charge of each state residing on the TCNQ (left panel) and TTF (right panel) molecules.

The TTF and TCNQ bands overlap at the Fermi level (1D metal)

The Electronic Structure of TTF-TCNQ

What's special for TTF-TCNQ?

The TTF and TCNQ bands overlap at the Fermi level

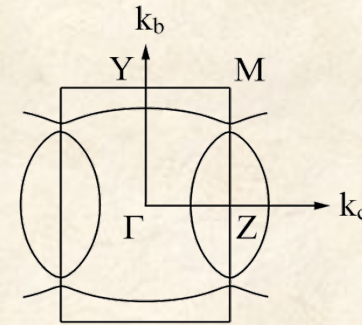
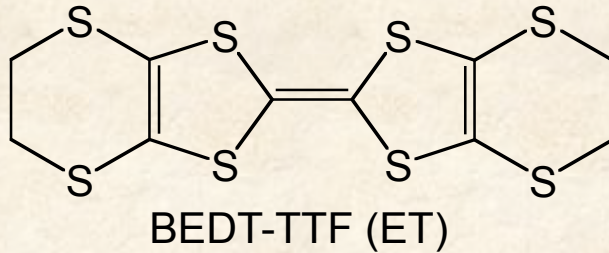


No distortion (at room temperature)

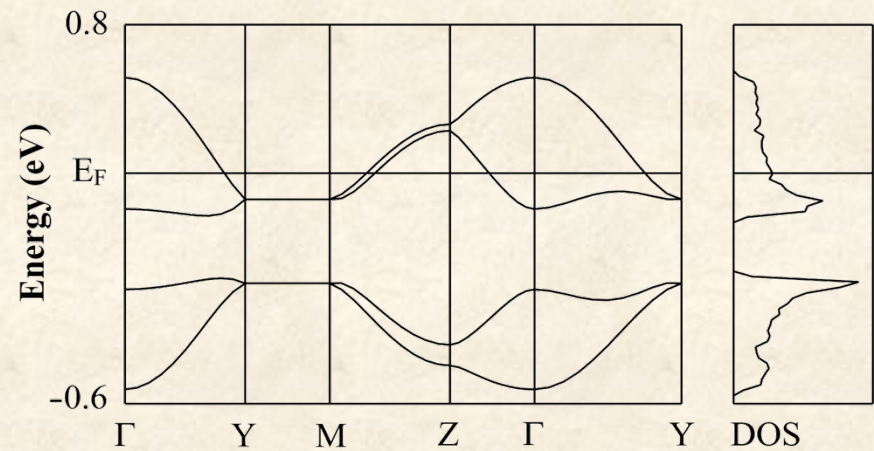
TTF-TCNQ is a 1D metal

Examples of Molecular Materials and their Electronic Structure

The superconductor κ -(ET)₂Cu(NCS)₂ ($T_c = 11$ K)

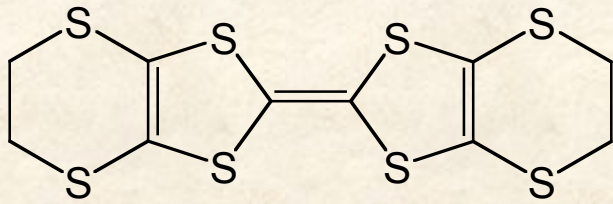


... An extensive work to shift the **Fermi level** by replacing part of Cu(I) with Cu(II) has so far been unsuccessful for κ -(ET)₂Cu(NCS)₂, however, successful in a very limited range in κ -(ET)₂(Cu^I_{2-x} Cu^{II}_x)₂{(CN)_{3-2y}[N(CN)₂]_y} [24]. A uni-axial strain of 1 kbar applied along the *c*-axis of κ -(ET)₂Cu(NCS)₂ increases T_c by 1 K [25] owing to the much **flattened 1D like Fermi surface** under pressure as expected from the **band calculation** giving rise to an enhancement of **electron correlation**. A uni-axial strain along the *b*-axis monotonically decreases T_c , probably due to the **increase of the bandwidth** and hence the **decrease of $D(\epsilon_F)$** .

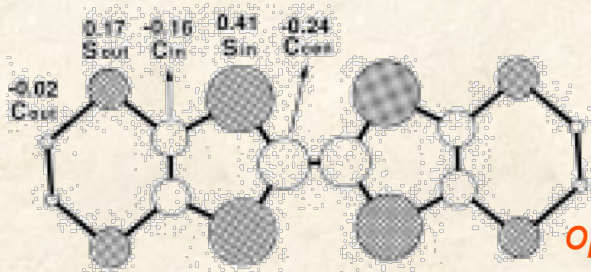


Calculated Fermi surface (top), energy dispersion (bottom left) and density of states $D(\epsilon)$ (bottom right) of the organic network of κ -(ET)₂Cu(NCS)₂

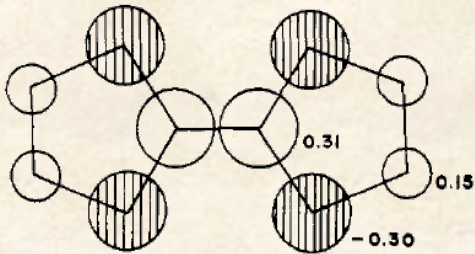
Electronic Structure of the Superconductor κ -(ET)₂Cu(NCS)₂



BEDT-TTF (ET)



HOMO of BEDT-TTF



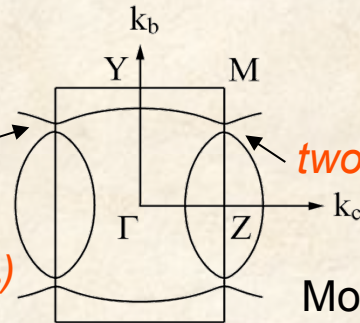
HOMO of TTF

4 molecules per unit cell

4 HOMOs (basis set)

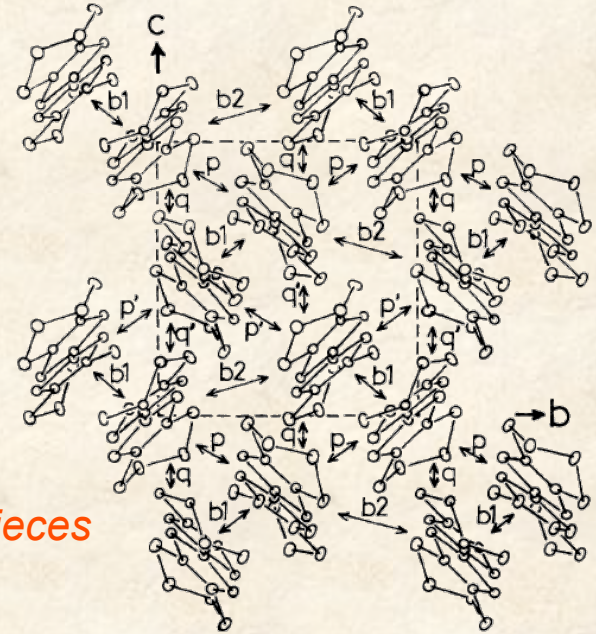
4 crystal orbitals (k)

4 bands ($E(k)$)



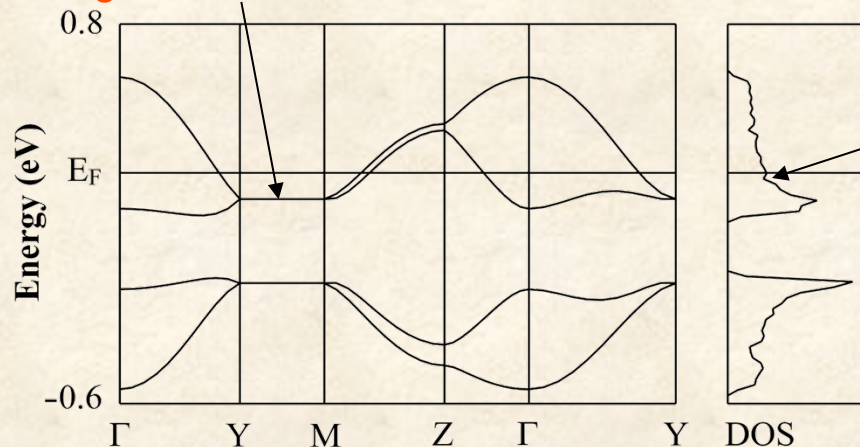
open Fermi surface (closed with I_3)

two pieces



Molecular arrangement of ET molecules

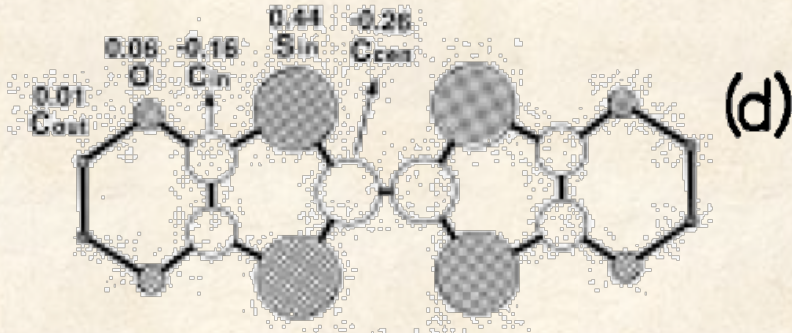
degenerate bands



metal character

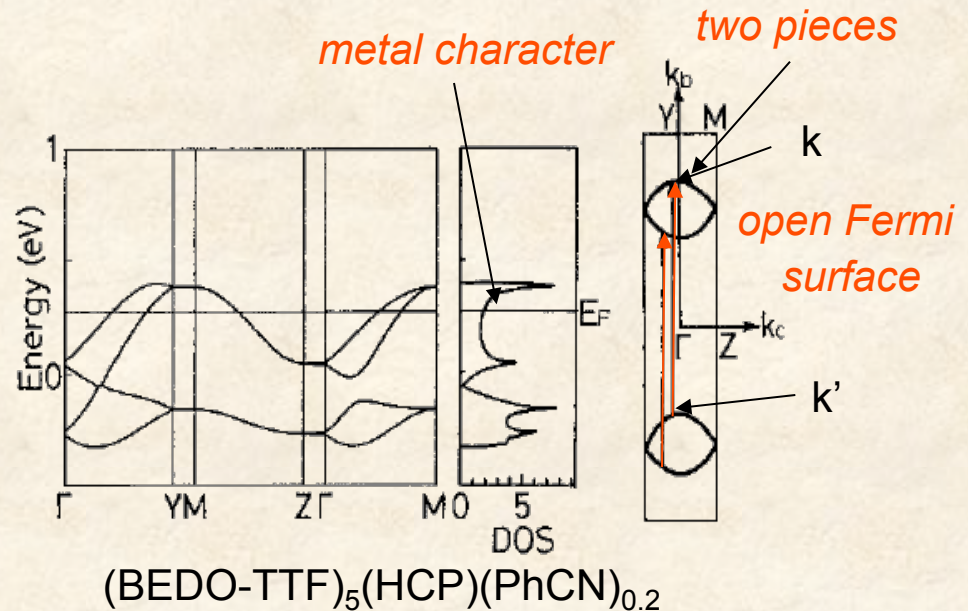
Calculated Fermi surface (top), energy dispersion (bottom left) and density of states $D(\epsilon)$ (bottom right) of the organic network of κ -(ET)₂Cu(NCS)₂

Electronic Structure of some (BEDO-TTF) Complexes



HOMO of BEDO-TTF

4 BEDO-TTF per unit cell



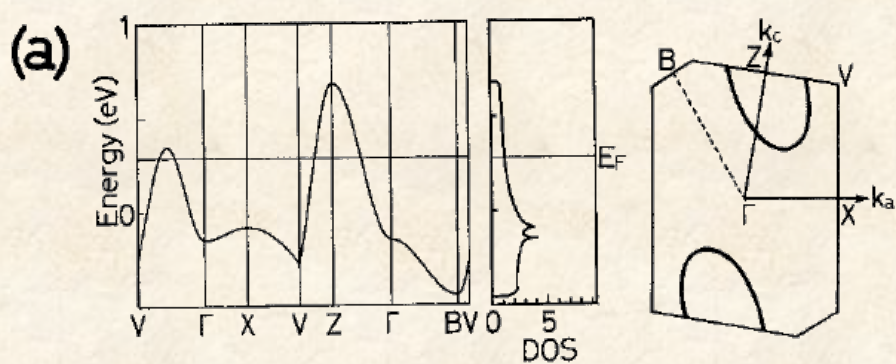
(BEDO-TTF)₅(HCP)(PhCN)_{0.2}

Calculated band structures, density of states (DOS), and Fermi surface of (d) (BEDO-TTF)₅(HCP)(PhCN)_{0.2}. The nesting vector \mathbf{q} is represented by arrows in d.

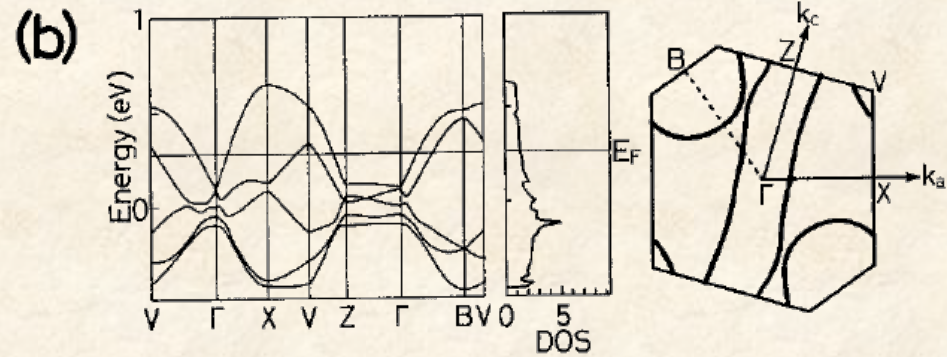
Two Fermi surfaces superposed to each other by translation vector $\mathbf{q} = 3/5 b^*$ ($\mathbf{q} = \mathbf{k} - \mathbf{k}'$ ($k = 0.3 \pi/b$)) (arrows in Figure).

This **nesting** of the Fermi surface (often seen in the quasi-1D materials) typically causes a CDW- or SDW-associated metal-insulator transition.

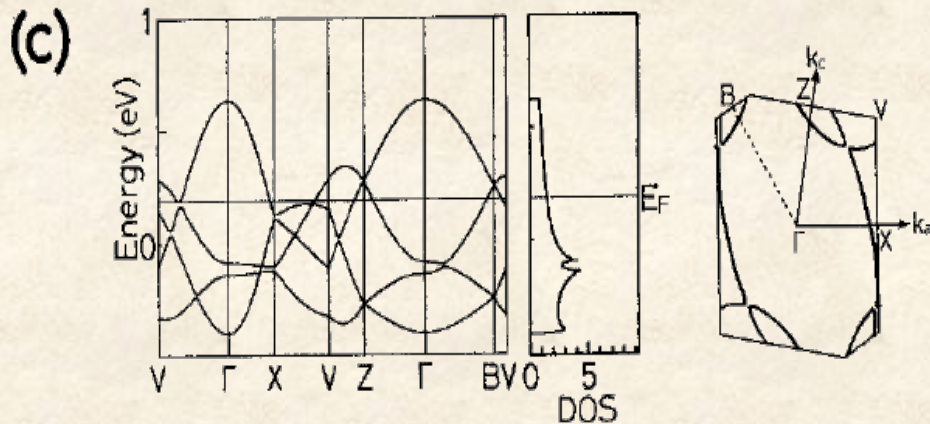
Electronic Structure of some (BEDO-TTF) Complexes



$(\text{BEDO-TTF})_{10}(\text{CF})_4(\text{H}_2\text{O})_3$



$(\text{BEDOTTF})_5(\text{HCTMM})(\text{PhCN})_2$



$(\text{BEDO-TTF})_4(\text{SQA})(\text{H}_2\text{O})_6$

*hole pocket around X (band nearly empty)
electron pockets around B and Z (band nearly full)*

Calculated band structures, densities of states (DOS), and Fermi surfaces of (a) $(\text{BEDO-TTF})_{10}(\text{CF})_4(\text{H}_2\text{O})_3$, (b) $(\text{BEDO-TTF})_5(\text{HCTMM})(\text{PhCN})_2$, and (c) $(\text{BEDO-TTF})_4(\text{SQA})(\text{H}_2\text{O})_6$.

⇒ *2D-metals*

Home Work: What is the Band Structure of $(\text{EDO-TTF})_2\text{PF}_6$?

Gigantic Photoresponse in $\frac{1}{4}$ -Filled-Band Organic Salt $(\text{EDO-TTF})_2\text{PF}_6$

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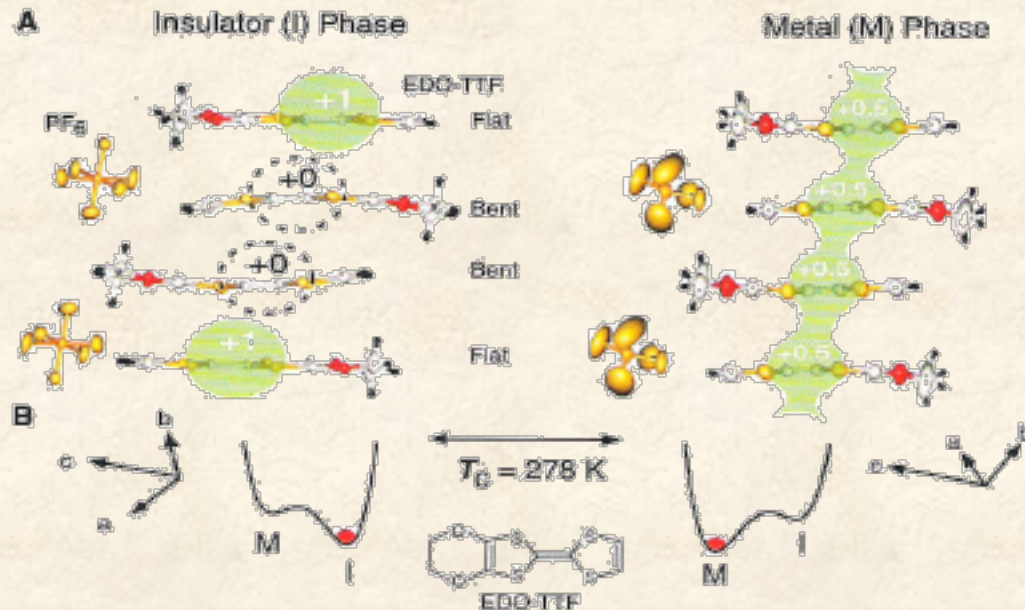
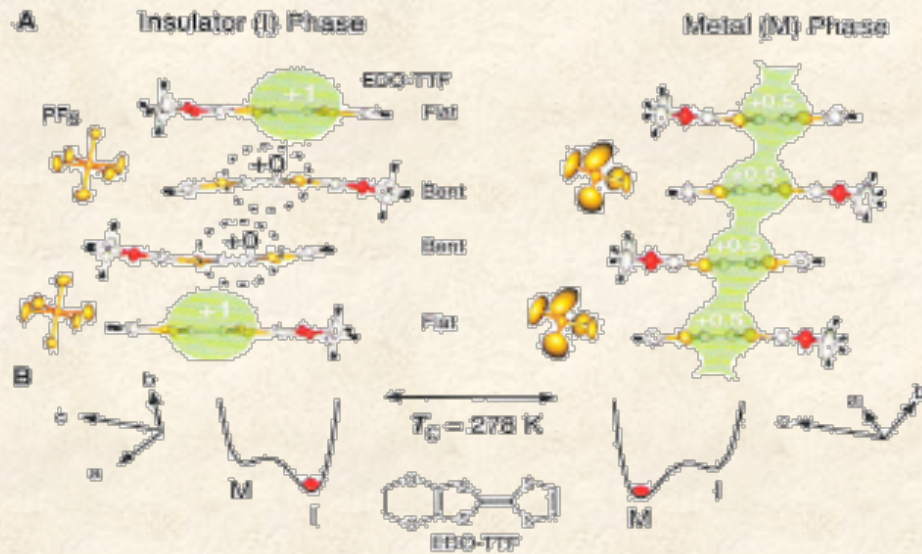
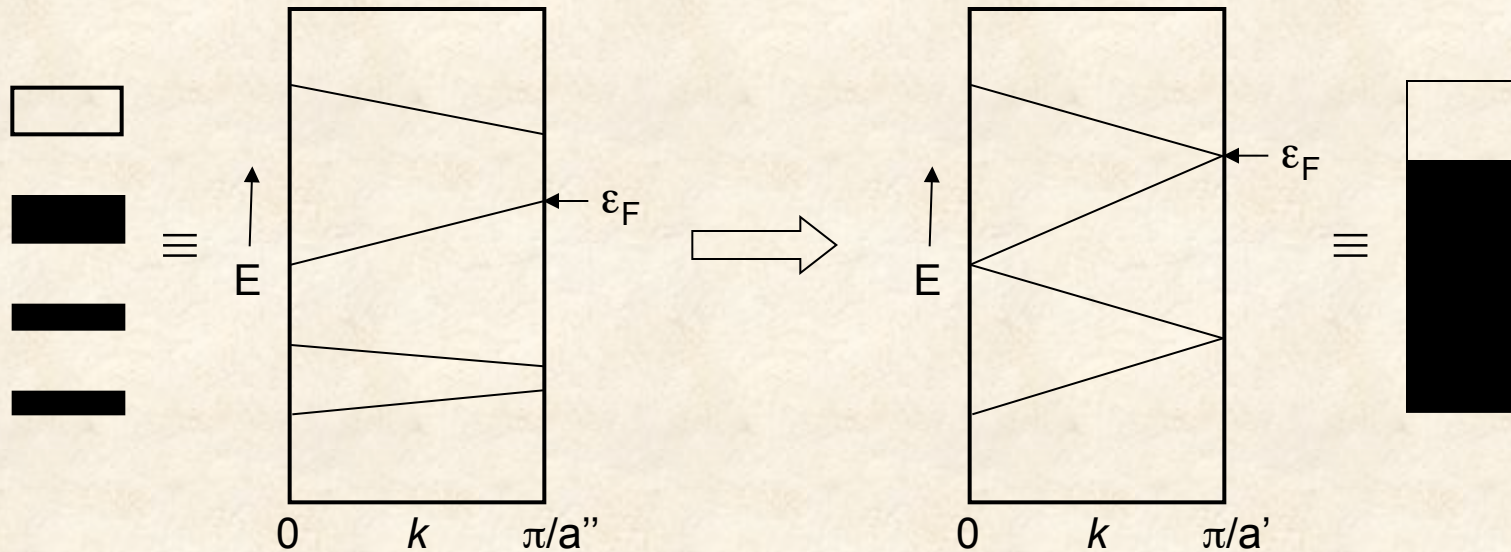


Fig. 1. (A) Schematic views of the lattice and electronic structural changes accompanying the M-I phase transition in $(\text{EDO-TTF})_2\text{PF}_6$. A side view of an EDO-TTF molecule is shown. The unit cell includes two and four EDO-TTF molecules in M and I phases, respectively (15). In the I phase, holes are localized on EDO-TTF molecules with a flat structure due to CO, and quasi-neutral molecules show a bent structure. In the M phase, charges (holes) are delocalized and PF₆ (acceptor) molecules exhibit disorder (15–18). (B) Schematics for the energy change accompanying M-I transition and the structure of the EDO-TTF molecule.

Home Work: What is the Band Structure of $(\text{EDO-TTF})_2\text{PF}_6$?



Expected band structure (four EDO-TTF molecules per unit cell):



Additional Reading

P. A. Cox, *The Electronic Structure and Chemistry of Solids*, Oxford University Press, Oxford, 1987.

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